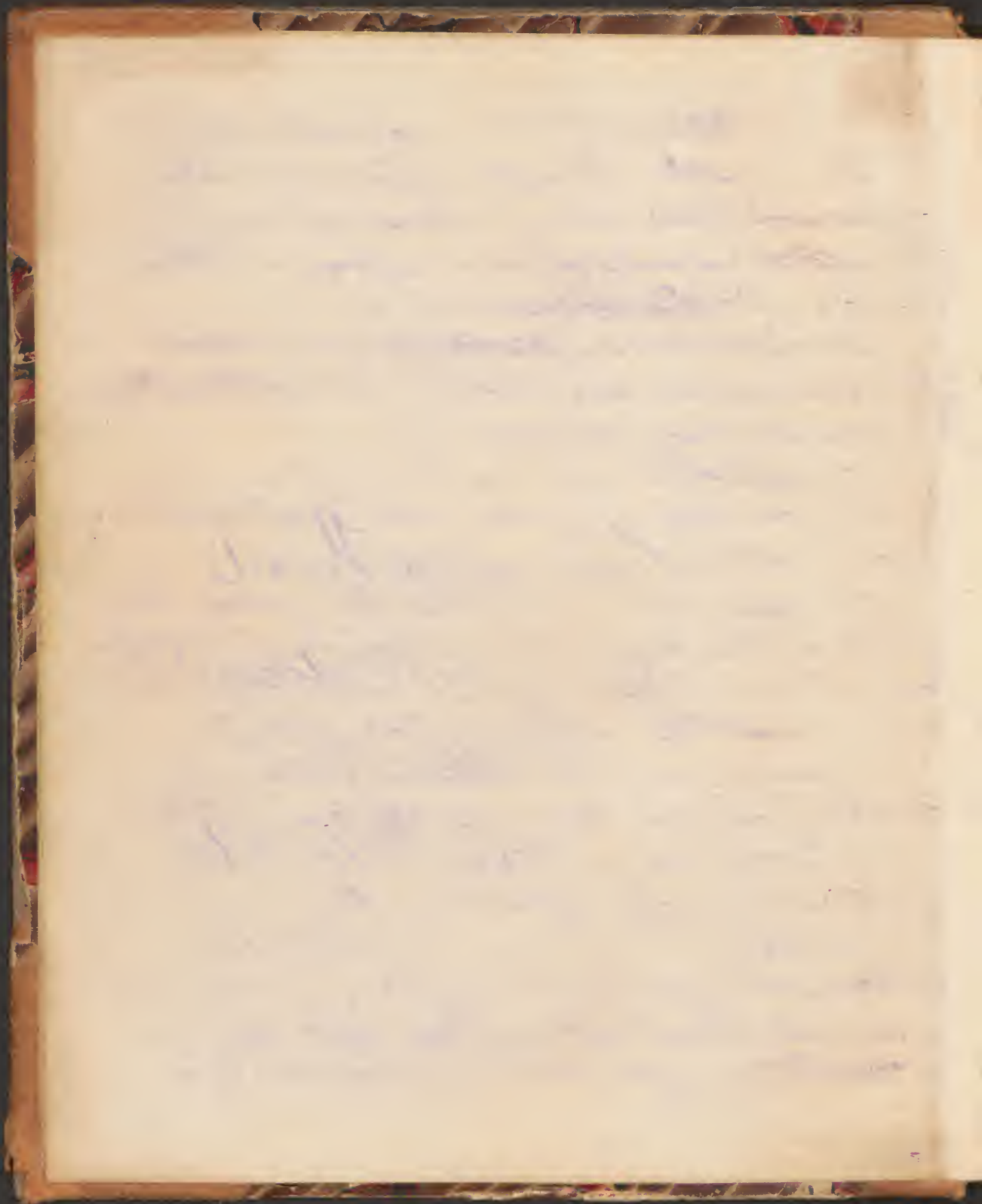


Gen. Leonard Wood's Notebooks at
Harvard Medical School Dec. 1881 - June 1882.

Leonard Wood
Harvard Medical School
13 Allen St
Oct 1881



Waterbury Med. Dec 15. 1881

Rec In the Phosphorus oil the oil is heated to remove any air or matter which may be in the oil.

Sulphur and sulphurous acid may be used for haemorrhoids, itching and ringworms.

Sulphur may be given in bulk as dose from a teaspoonful to a Table spoonful.

Sulphurous acid dilute may be given in doses from 10 to 30 drops in a tumbler of water. Strong 1-3 drops to the same amt of H₂O.

For treating the trachea in better than the compound tincture.

The latter is better than the pure tincture for extirpation.

The tincture is used in water. The compound tincture is 200. in water and the volume is 200 in a solution of potassium hydrate.

Iodine is given internally in the
form of a tincture very dilute but
even then it is very irritating
and is given in doses of 20 grains

It is also given mixed with
starch which makes it much
less irritating and may be given
in larger doses 4-6 grains

Doses of Potash may be
administered in 2-3 grains which it
takes very well

It is given with a mixture
of Rosaphyllin which
helps all the symptoms

Potassium Hydroxide. It is a
caustic and acts on a surface and
is given externally as an antiseptic

1

(3)

(4)

(3)

(4)

(2)

(4)

(2)

(178-216)

80

66)

(1)

(1)

(1)

(1)

(1)

(1)

26 - 2 (1)

(2)

88

alcis

84

55 - 1 (75)

85

1.5 - 1 (50)

83

3 - 1 (5)

1.5

26 - 2 (1)

H₂O

H₂O

The water is pure and clear
and is collected from the
the air plants, etc. in some cases
the water is used for the
purpose of watering the plants
and for the purpose of
watering the plants.

Polypodium autumnale, the
a mass of several clusters of
fronds, collected in the garden
of the University of California, Berkeley, 1890.

The specimen here is a mass
of fronds, autumnal and
spring, collected in the garden
of the University of California, Berkeley, 1890.

Polypodium autumnale, official name
H. 11.
The specimen here is a mass
of fronds, autumnal and
spring, collected in the garden
of the University of California, Berkeley, 1890.

This is a small country, but a
very good one, about 100

There is a good road on
the coast, a narrow one, but it is
the only one, the best of the coast.
The water is very good, but it is not
very good.

There is a good road, but it is not
very good. The water is very good, but it is not
very good.

The water is very good, but it is not
very good. The water is very good, but it is not
very good.

The water is very good, but it is not
very good. The water is very good, but it is not
very good.

The water is very good, but it is not
very good. The water is very good, but it is not
very good.

The water is very good, but it is not
very good. The water is very good, but it is not
very good.

...with the ...
...it is ...
...the ...

...to a ...
...the ...
...the ...

...the ...
...the ...
...the ...
...the ...
...the ...
...the ...
...the ...

...of the ...
...the ...
...the ...

Yellow sulphur & oil of turpentine

Being of different of the same
and also being of the same
nature & being of the same
nature of oil

There is a great amount
of the same of different
nature & being of the same
nature of oil

It is of the same nature
of the same nature
of the same nature
of the same nature

It is of the same nature
of the same nature
of the same nature
of the same nature

It is of the same nature
of the same nature
of the same nature
of the same nature

L.

Antimony	180
Sulphate of antimony	"
Sulphurated ..	181
Fluorine Hill	1094
Crude of antimony	180
Fluorine powder	1184
Tartar Emetic	183
Opelaitum antimonii	528
Vinum antimonii	1522
Distillate of powder of tartar	925
Syrup of squilla	1401
Oil	
Sub-carbonate of tin	288
" Nitrate ..	292
Tartrate of tin	293
Hyd.	735
Carbonate of tin	742
Hydrogen sulphide	
Fluorine of a iron mostly	
Yellow sulphate of Hg	733
Amellin	734

- Cent of nitrate of Hg
Boracide and Br sulfide of Hg
Oz Nitrate
Z
Z - Sulfate
Cu
Chloride

Materia Med. Jan. 5. 1862
Prescriptions are never written in Latin.
In using the Metric system the
Roman or Arabic numerals are
used.

Prescriptions consist of two independent
sentences, 1st. that of what
is to be given, & the purpose, & 2nd. that of which
I give is the
predicate.

Avoid all abbreviations.
Clearness of construction.
Every drug should have a line and
no more.

It is a good idea to give the name
of the patient.


It should sometime be written under
the signa. respect when two or more
prescriptions are going to the same home.

Age of the patient, female or male, child
should be set down.

Name of the prescriber should be
put on the last prescription.

When the dose is of great importance
something should be used to indicate
that it is not a mistake and
that has attracted the attention
of the ~~physician~~ doctor. In Germany
the ! is used here the word !
is used, but it is best to write out
the direction in full.

Modena Medico, Jan 10. 1882.

 Rules for doses. They should
be proportionate to the wt. of the patient.
This is Clarke's way. Another way is
to divide the age of the patient by
12 plus the age of the patient
(age). In very old persons give
(age + 12) have to diminish.

Amount of the prescription you
should bear in mind the amount
of the sub in proportion to its dose
and the length of time it is to be
taken. Notes Alcohol. fixed oils. glyce.
syrup. These are the best & easiest
all of the deco and infusion
will spoil in a warm room in from
3 days to a week.

Imdancer keep making. Squared also
Keep well a month or so in an-
nother

Matured March Jan 26, 1882
Starch this is formed in almost
all plants, as a partial energy
product, and is used in the first
for the plant and the proper

Starch formed in the cells and
is found in various kinds
of the plants in the process of
refining it disappears.

As long as starch is found and
good there is no doubt that it is
little difference of what is noted it is

There are large oval granules and
then small ones not over 10
in length, and some intermediate

are forms. Their size is $\frac{1}{500}$ of an inch
in all well organized gran gran
have a black spot in each end
Rice. starch angular.

Starch is veg. slightly 200 in old
+ 120. In fact it is a question

whether they are offered
in powder or in boiling H₂O
boiling at 200 deg.

Amylum Amylum or starch
several compounds of starch are
used in med. and pharmacy
Glycerinum amyli. 1 part starch
8 of glyce. used oil for eye wash
and used to hold other sol
solutions of starch, is a mix of
starch and I the I being - small
amt.

Good as an antidote in cases
of poisoning. Arrow root. this is
quite an important prep used
to be used quite extensively
as a food for chil. comes
from a root grown in trop
countries. close gran border
separating seen between the fin

Can be starch another variety
of arrow root
This root is used - richer
of I ~~part~~ ~~part~~

Gum. 2 var. 1 called *arabum*
2nd in H₂O 2 *valolin*
Scabiaria is one of the more
large. found in Trop countries
this exudes as tears and is
and is plucked in this form
Turkey gum then - the best
called *cordifera*

Gum Arabic is composed of *arabum* & *calcium* &
magnesium - Arabic acid readily obtained from
Gum Arabic - The Gum Arabic is used
in lozenges and pills - but always mixed with
glycerine and water owing to its hardness. *Tracilago*
used as a vehicle -

Tropicau - found in the wild of Europe
and Asia - A few.

Antimony

190

grm

Pilulae antimonii composita^e or Plummer
 Kille = $\text{St. S.} + \text{H}_2\text{Cl} + \text{Gum} + \text{molasses}$ 1094 1-2 Pills

Pulvis antimonius, or James powder. 12-1.
 = Oxide of An. + Phosphate of lime 1186 ~~12-1~~

Antimonium et potassii tartar, or tartar emetic
 = $\text{St. C.} + \text{B. tartaric of K} + \text{H}_2\text{O}, (\text{H}_2\text{O})$ 183 106-.03

Squippus scillae composita^e. = Tartar emetic + squill
 + seneka + Sugar + dilute alcohol 1401 ~~12-2~~

Mistura glycyrrhizae composita. Compound
 mixture of licorice = Licorice + sugar + acacia +
 compound tinct opium + Vinum antimonii + Sfr of
 nitrous ether + H_2O 925 16.

Bismuth

294

Bismuthi subcarbonas = $\text{Bi} + \text{HNO}_3 + \text{NH}_3/\text{H}_2\text{O} +$
 $\text{NaCO}_3 + \text{H}_2\text{O}$ (HNO₃) 288 16-4.

Bismuthi tannas = $\text{BiNO}_3 + \text{HNO}_3 + \text{NaHO} + \text{Tannin}$
 (HNO₃)

Hgdrargyri Pilulae. = $\text{Hg } 384\text{gr} + \text{Confec of rose}$
 570gr Licorice root 192gr make 384 pills.

1199 12-4

Hgdrargyrum cum creta
 cum creta = $\text{Hg} + \text{chalk}$.

not suitable for use as drops. can be used for
 742 0.2-.6

Prep dose

- Hydragryi sulphuretum rubrum, or ter-
- mition = $Hg + S$ mixed Sol in aqua regia used
as a fumigant and disinfectant 735

$Hg_2Cl_2 = Hg + H_2SO_4 + NaCl$ 779 Sol

2 part boiling 10 of cold H_2O 719 .004

Hg_2Cl_2 Hydragryi chloride = $Hg + H_2CO_3$
+ $NaCl$ Sol in the mineral acids. For,

used given in a syringe in water. 722, .06-1

Hydragryi iodidum rubrum. $Hg_2I_2 + I_2 +$
 H_2O Sol in hot alcohol in a
powder. 720 .004

This latter is called the Biniodide of Hg
 Hg iodidum rubrum = $Hg + I +$ stronger alcohol
insol in water and alcohol. 730 .55-56

Hg sulphas flavus = $Hg + H_2SO_4$ 733 .2
Zinc 1550

Zinci sulphas = $Zn + ZnO_3 + Cl + H_2SO_4 +$
 H_2O , Sol in a large excess of H_2O 1547, ?

As an emetic it is given dissolved in water.
1520 in a dose of 5-4.

Requires dose for continued treatment .06-1
in pill form

Cu 485

Cupri sulphas = Sulphide oxidized in the
sulphate Sol in H_2O 482

Amomum 12, 1-6 mixed with full
sugar; Continued treatment - full form .01 - .1

Amylum 10 of 8 to 1 of 1 171

Iodidum amyli 10 of 8 to 1 of 9 776

Glycyrrhizum amyli 10 of 8 to 8 of 9 686

Tragacantha, socialem per the tree astragalus
cens. formed in bunch 1 wide and 4 or 5 inches
long. of a white color 1465

Acacia 3

Specacanthu, opium ^{4 gm} 60 gr each. KBo4 ^{8 gm} 32
rub together into fine powder, = Comp Specacanth 1188. (1)

Pulvis glycyrrhiza compositus. =

Powdered Lenna, powdered licorice root, each 60 gram
Refined sugar 180 gram, make powder 1188 (3)

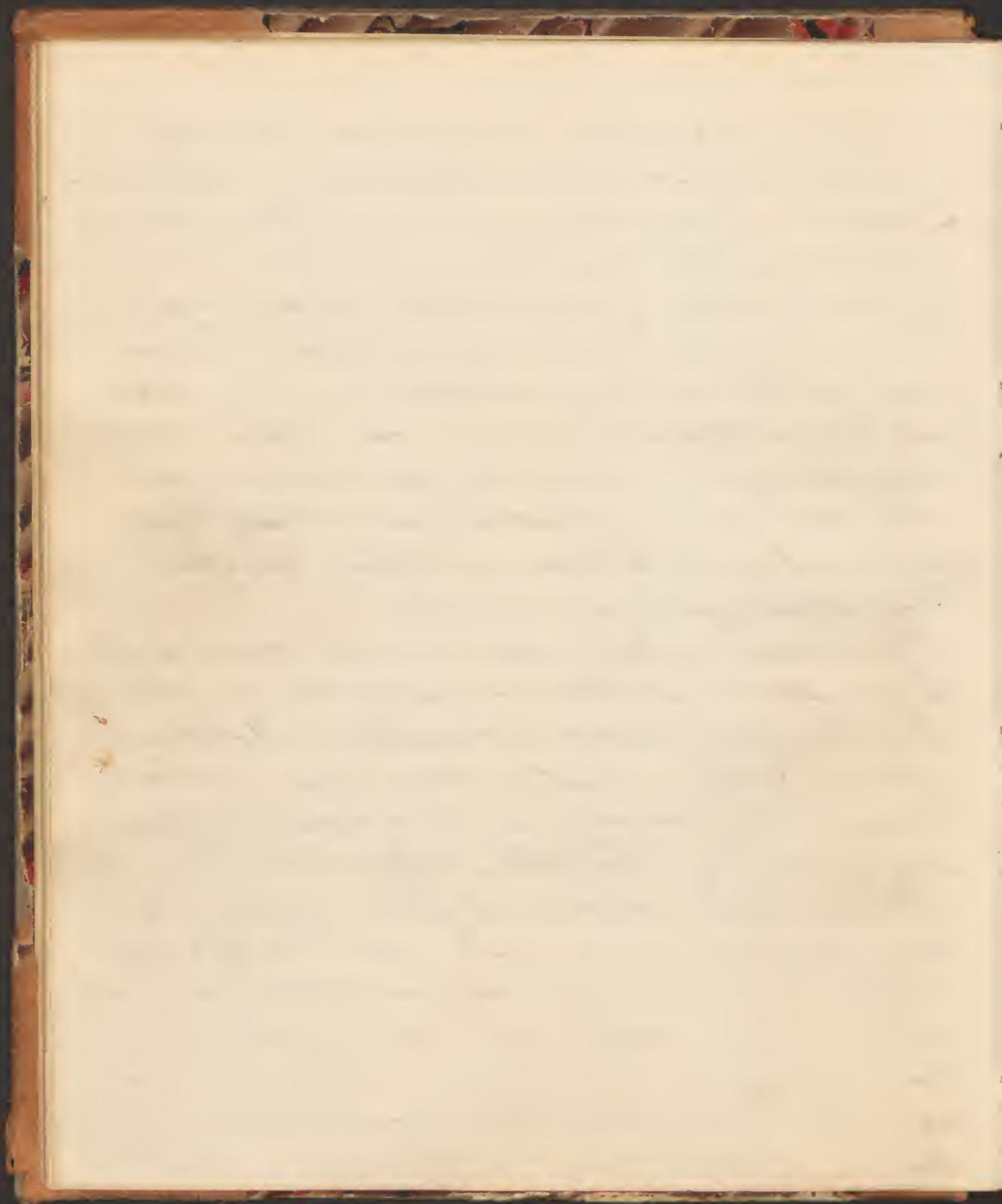
Pulvis jalapae compositus, = Jalap -

powder 30 gram. Tinct of potassum 60 gm. 1189 (1)

Pulvis opii compositus = $1\frac{1}{2}$ oz. of Opium, black pepper

2 oz. ginger 5 oz. ^{opium 6 oz.} ~~Max 1189~~ 1190 (2)

Pulvis rhei compositus = R. Tinct 4 oz. Mg 12 oz.
ginger 2 oz. 1196 (2)



Material Med. Feb. 1882

Lecithin, Manna, Glycerin in the sugar
which is the important portion of the Lecithin
Stick Lecithin has been made by boiling
the root down in a thick syrup and
filtering and ~~season~~ rolling into
sticks. The syrup of lic is very use-
ful as a ~~vehicle~~ vehicle: as in the powder

Manna comes from the bark of an
ash tree grown from in ~~early~~ early
and southern Europe. ~~Early~~ ~~Early~~

Cellulose is used in med in the form
of ~~granulation~~ granulation. This is heavier and darker
than normal cotton it will dissolve
ether and alcohol forming colloidal

Used to cover lacerations and to fasten
Also used in holding wounds together
As it dries it contracts strongly

Wood exposed to a light heat it becomes
porous into spongy and one of the most
useful in the. It can be used as a
vac. sol in ether alcohol or chloroform
usually given in form of an injection
made by macerating 1 of the in 4 of
water. ~~ether~~

Glyce of tar, oint of tar
Infusion is given internally as
many the glyce but not as much
Oil of cade. another form used
by dermatologists. sup. to be made
from cedar.

This looks like tar and has the
prop but the odor is more agree-
able. Creasote made by distill of
tar, and sep by alk. is of a yellow
ish. or reddish color which it gets
upon standing. Sol in 80 parts of
H₂O. Sol in ether alcohol.
good antiseptic, used like
Carbolic acid. Made as a oint
and as an ointment
1 lb 80 in the lowest amt of the
which can be used

Charcoal. Carbo ligni. or tar
is made in this country in a
very crude way.

It is used as a deodorizer
and is applied to elongated sores
is given internally in cases

where decay is from wood. The
food putrefies in the stomach.

Paraffine made from wood and
a coal tar, comp of several sub
Insol in the slight alcohol
used in a mix in the harder
eco cements.

Carbolic acid made in the
distillation of wood and
coal. rarely comes pure, is
crystalline at ordinary temp
when pure. Cryst are separate and
distinct mass of a white color.
Melts comp of today. absorbs
H₂O on exp to atmos phere.

Sol Mix of 5 grs of H₂O is not a
true Glycerate, 1-4, Suppave
Oint. Sol in water. These are
the forms which are used.
Salicylic acid is made from
Carbolic acid, and is much
sharper than the old form which
was made from winter green
Cresoline base line

	Page
Squarice =	580.
Collodium	443
Tur	1109
Oint of tur	1493
Effluents of tur	687
Charcoal	349
Cassidi	471
Carbonic acid	31
Squarice Sulfuric acid	73

Materna Medica Feb 9, 1882

'Salyntia' Salicylic acid given crude and
in case of rheumatism given - powder and
in solution. dose 25 to 1. gm. Taken
in powder with a little something to
deguise its taste. When given in 120
grs should give it in an alk
water. Salicylate of Na. in a little
more sol. than the common variety
Carbolic C_6H_6O . It is also given in
liquid form. when pure it crystallizes
in the form of pure soft decahydrate
crystals which melt at 20 deg. when
mixed with glycerine & water it
is much more convenient
A teaspoonful to $\frac{1}{2}$ pint, in a good
proportion, 1-20 is the strongest that
can be made. 1-40 or 100 is done
as strong as need - in grease.

In oint 1-16. is right.

Petroleum in the crude state has
been used externally and inter
but is not of much use

Cosmoline Paraffine. Cosmoline
is the end left in the still

after the distill of the more vol
sub. Pyoline is a sub which
comes off during the first portion
of the distillation is used as a
means of local anæsthesia

Neither benzene, vasoline are the
sub which come over after pyoline
Next come the paraffines which are
waxy solid in mass in it is not
sol in hot alcohol

If the distill is not carried far
enough to carry over the paraffine
you get a dark sub with a
strong odour and when this is
passed through animal
charcoal, and you get the
clear comoline and cascade

The charcoal is heated just
enough to cause the oil to come
over through in a liquid state

Comoline never crystallizes
and have this advantage over
paraffine

All these sub never become rancid
and are thus better than the animal

fat. and they do not decompose
the drugs with which they are
mixed

Sub-made by fermentation
1st Alcohol - When beer etc they all
depend upon the fermentation of
sugar. in presence of a little albumin
or sol. ~~with~~ with the presence of heat
There are also a certain number
of subtle odors sub called ethers
which give a certain flavor to the
liquors.

One medicine made from malt
extract. not officinal - malt treated
treated with H₂O to remove old
sugar. and then evaporated to
a syrupy consistency contain grape
sugar dissolved in H₂O. and a little
amt of diastase not very valuable.

Other products made from malt
are the beers. ~~as~~ as alcohol 4-5.

Ales. 6-8. porter and stout 6-8.

Porter and ale colored by addition
of caramel

Wines are made from such fruits as do not need the addition of yeast. but have a medium in which the sugar fungus takes root

Wet and dry refer to amt of sugar. White and red refer to the color of the fruit

Sparkling wines are those ~~which~~ bottled before fermentation stops and so you get a little CO_2

Officinal wine are Port and Sherry Port is a red wine with a large amt of tannin and comes from Portugal
S. I. ought to have 20 or 22 %

Prun wine is lighter and does not contain so much tannin
Usual has from 16 - 20 %

Rome wines are prob more pure as are the lighter French and German wines generally. They usually have a good deal of free $\text{Bi Bi tartrate of K}$.

Alcohol distilled from wine make brandy

That distilled from grain muck
or ~~brandy~~ Whiskey. Molasses makes
rum

3 grades of alcohol on the phar
stronger alcohol = 92% of alcohol
ordinary alcohol = 82%.

Proof Spirit = 50% of alcohol
Alcohol is used in the fluid ex-
tractions, resins and others

From this comes ether, by distilling
alcohol with H_2SO_4 . That which comes
over between given temperatures, it
is a clear volatile inflamm sol
sol in H_2O in proper of 1-18. or
if a little alcohol is added it
is more sol

Mixes with alcohol sol to any
extent. fats, volatile oils are all
sol in it. as are the greater part
of the alk. acids

Ether alcohol & ethereal oil

Used for conflict and local
anesthesia. But for the cure is
not good. Internally dose is
a tea to a table spoonful

best way is to enclose in a cap
 sure. By the return this is good
 for colic. should be mixed with
 starch or some such sub.

Prep are the compound spirit of
 ether, and spirit of rutine ether
 3

Subjects for examination recitation

Maranta.		899	
Acacia	Ley.	3	
Traciantha	Ley.	1465	
Glycyrrhiza	Ley.	688	
Manna	Oleaceae	897	
Saccharum lactis	(Sol in H_2O)	1246	
Collodium	(Sol in E. & Alcohol)	443	.05-1
Crotonum	(Sol in H_2O , then alcohol)	471	.15-1
Acidum carbonicum	" " " "	31	
Cosmoline	(Sol in oil and fixed oils)	1055	
Paraffinum	" " " "	1057	
Whiskey		122, 1343-	
Hine		1521	
Cum		122	
Brandy		122, 1350	
Beer		7, 618	
Alcohol		121	

Amyl nitras (Sol in rect spirit)	168	.1 - .3
Ether	105	
Spiritus aethereus compositus	1338	.2 - .4
1. Nitric aethereus	1339	1 - 2
Iodoformum Soln (C_2H_5O)	780	.06 - .2
Chloroformum	401	1 - 4 Sub cut
Chloralum hydratum	393	.6 - 2.

Also read up the propyl gums and starches
 L L L L 4

Materna Medica medica
 Feb 16. 1882. Recitation ~~Feb 16~~

Best typeantha is called flake tray-
 ionic

Sub ject for recitation Mar 9, 1882

Oleum aethereum	977	
Spiritus aethereus compositus. ()	1338	.2 - .4 (3)
" " Nitric	1339	1. 2. (1)
Lomentum chloroformi	828	
Mustina "	922	16. 32. (20)
Spiritus "	1344	1 - 4. (2)
Amyl nitras	168	.1 - .3 (2)
Buchu Rutaceae	305	4 - 16. (10)
Oleum menthae piperitae	998	.06 - 2. (1)
Spiritus "	1348	3 - 1. (5)
" "		16. (10)
Oleum camomili	990	.06 - 1. (1)

Materia Medica Feb 23. 1882

Internal use of Chloroform.

It is used as a solvent for many sub
only one officine. Chloroformi guttae butcher
Drop on a lump of sugar in a good way
to take it dose 10 drops

It also comes in capsules of gelatine

Ex temporary it may be mixed with
any of the viscous fluids. There is
a white sol of Chloroform - 17.5
1000.

Ether is used in various forms its
white of amygd is another prep of the
does urine given by inhalation. and
it causes flushing of face and rapid
breathing. and causes a cessation
of unimpeded muscular. After

3 inhalations are enough to start
with. It is a good plan to tell
the patient what the effect of the
med will be. Given internally
dose of 3 drops

Chlorine in presence of the air is
slightly deliquescent.

It is given dose in H₂O in which it is set
to any extent frequently given in syrup

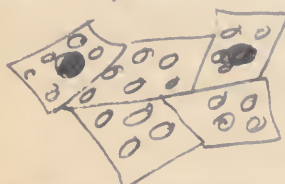
Has a flavor like rancid English
walnuts should be largely diluted
with H₂O. given subcut. but then
painful. best way is to give it
internally. dose is 1 grm. It is some
what dangerous. medicine and
should always administered with
caution. Good way to give this is
to put each ~~way~~ dose in a separate
bottle and sealed up. and thus it
keeps good and is always

Iodoformum. mol of H₂O. 200 in Ether
oleat. *Chloroform*. Used externally as
a stimulating oint. and also given
internally it is stimulating. best given
in the form of a pill.

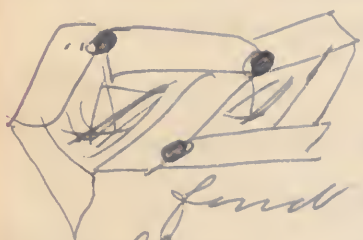
Dose for internal admin. is a decigram

Essential and volatile oils
Champhora resin, *gum resin*
and *balams*

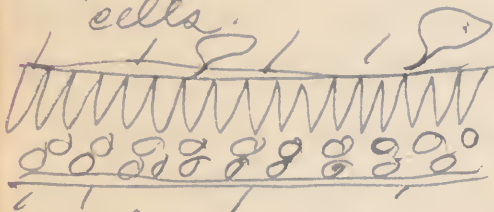
Essential oils are usually already formed vegetable products, and are insoluble in the juices of the plant, and occur as separate and distinct masses.



Appearance of ginger under the microscope showing the starch granules and in every few cells we see the refractive masses of the essential oil.



Longitudinal show the cells of sweet flag, and here the oil globules are found at the junctions of several cells.

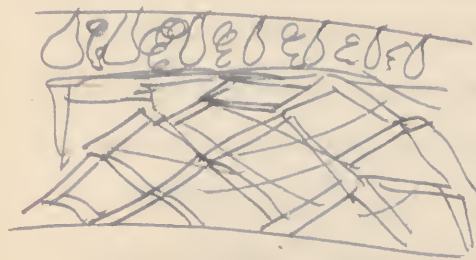


Showing the position of oil globules in the rays.

In some plants the cells of essential oils are in a capsule with a long pedicel.

In red Buchu and orange juniper leaves you see the cells of the essential oil with the naked eye.

Cell on the outer end of orange
 appearing on the outer
 layer of the rind the
 inner two thirds
 being of fine connective
 tissue



Cut through a piece of
 pine wood, showing oil
 cells. In the parsnip fruit the
 oil cells are found.



Cut showing the latter



In some of the fruits the oils are
 found in the form of emulsions
 such as gamboge ~~the~~ tree.

Prop of these oils. There are not
 usually found singly but you also
 have resins mixed varying
 only in their densities and
 boiling point and then varnishes
 with prolonged light

This is the case with the oils of turpentine, and are separated by careful fractional distillation. Then again they are associated with oxygenated oils these are separated by the action of deoxidizing agents. The one thought to be formed by decomposition of essential oils - resins are associated with these due to above mentioned degen.

Essential oils are pure white mobile oils and the longer they have been distilled the darker they are.

Some oils have a greenish hue when first distilled such as oil of Camomile.

In some cases they are of a bright blue color. These belong to family umbelliferae.

Sp gr varies from 300 to 1000 more than $\frac{9}{10}$ are below the density of H_2O .

When oils are of diff density ~ ~
the same plant the best odor
is obtained from the lighter
Altho the oils are sol ~ H₂O
to extent of 1/1000. All sol ~ Ether
alcohol and chloroform. They
can be distilled and redistilled
without change.

They are usually separated from
the dregs contain, then th by distill
ation. In orange peel by pressing
the rind upon sponge. In the
the prep of perfume they are obtaine
ed by peeling ~ got.

The oils prepared by pressure are
above the most fragrant

Orange family oils are from
Citrus aurantium. = oil of orange. for
rind. - essential oil for leaf. used
in perfume. an oil prepared from
the flower

Same ~ the same for yelder
the same.

Bergamot = oil for rind
Citrus, also given in oil

Most family. give a series of
volatile oils. Lavender. pepperment
Sassafras, rosemary sage. are
in this list

Bucher pilosus.

Umbelliferae. = Caraway anise
Coriander. 5 N.O.

Subjects for rec. Mar 9. 82.			
<i>Tractina cinnamome</i>	Lamiaceae	1441	(4)
<i>Aqua</i>	Labiatae	217	
<i>Spiritus Loomaiae</i>		1347	(3)
"	Comp	1847	(3)
<i>Thymus</i>	Lamiaceae	1023	(2)
<i>Spiritus camphorae</i>		1344	(1)
<i>Aqua camphorae</i>		214	(1b)
<i>Liramentum</i>		827	
" <i>Opium</i>		835	
<i>Tractina opus camphorata</i>		1454	(5)
<i>Cubeba</i>		477	
<i>Oleum cubebae</i>	Leg	975	(1)
<i>Tractina cubebae</i>		1442	(4)
<i>Fructus Capata</i>	Leg	459	(5)
<i>Coposic acid</i>		460	
<i>Oleum Coposae</i>		991	(5)

Materna Medica Mar 2. 1882

Camphor is an oil solid at the ordinary temperature of the body. another is tyros and methoc is another these melt readily and then have the appearance of the ordinary essential oils. Stearoptene is the name applied to these. They are associated with a hydrocarbon and have the formulae of the hydrocarbon. These have great antiseptic prop. They are 200 - 1000 parts of 72

Sol in all prop of alcohol.

1 - 1000 is right for an antiseptic

Resins are decomposed of essential oils and are very abundant. they do not resemble these. They are mixed in H₂O. has an odor of the oil usually are transparent, may be amorphous or crystalline. Sol in either alcohol and ether some are sol in one of these and some in both. ~~are sol in~~ some are sol in alk forming soaps.

Gum resins are made out of vol oil large amt in resin. usually more than one half and they also have a gum

2d - H₂O. When found in the plants they are in the form of an emulsion usually white or yellow color.

The resin is not volatile but in a state of fine subdivision in the fluids of the plant.

Balsams there may then be liquid or solid, all have essential oil and a complicated list of resin substances consisting of crystalline and amorphous substances among which is benzoic acid.

Myrtle family gives oil of eucalyptus. has a green color, this is quite distinctive. *Eucalyptus globulus*.

Uentiptus globulus this is our Australia tree and gives two kinds of essential oils, and the lighter of these is a pure hydrocarbon.

Leaves have a curved edge when a cythe, *Eucalyptus globulus*.

~~Myrica~~ *Myrica* *Myrica* gives the oil of bay and you also get the pure bay resin.

Cloves. These are dried flower buds and are native or are found in south
in Asia Brazil. found of size of
an ordinary pear tree.

The tree is beaten and these little
buds drop off and are gathered
and dried and sent into com-
merce. On rectum the great size of
the oil cavity is remarkable and they
have the largest % of oil of any crude
drug in the world. It gives a
pure essential oil heavier than
oil water and upon distill it
gives a genuine oil.

Allepree or pimenta. This is a
West Indian drug: gives an oil
called oil of allepree and is a little
colored. This forms the myrtle
family.

Mitney tree. This is a family
to itself native of India.

Has a fleshy outside layer out
of which comes the richness of com-
merce.

The seed and of the fruit produce
mace. This mace producing matter
surrounds the seed or nutmeg
and comes out with the salt
and is separated after ward
Oil of nutmeg is obtained by pressure
and has a large amt of solid
fat, composed of palmitine and
other fats.

It also has an essential oil
which has the characteristic odor

This solid fat is known as oil of
mace. Though mace has none of it
Lamel. - acacia acacia cham
phor. sasaparilla. common.

Campor. is prepared by putting
pieces of the wood in a still and
steaming and the camphor comes
out in little drops. at the time
an oil also exudes called the
oil of camphor. it is an essen
tial oil

Common. bark of cinnamon
cinnam. cinnamon flower.

Ceylon, There is a tree growing
in the south of Asia which is
not common, and is curled on
fellows and is thick. (C)

True cinnamon is thin and curled
in at both ends and is composed
of various layers. (C) is an almost
solid stick

Siam, there is found above the
northern and middle state and
northern, the bark of the root is
the part used

Essential oil comes from the
the whole tree.

Pepper family. Piper and cubeb.
are most import, Both are the
fruits of the plant and are: perian-
them because they have more
oil and resin. Pepper is black-
color. Cubeb are of same size and
have a long stem and the
long stem assists in recogni-
tion of the fruit.

The stem is a prolongation of the

celia. White pepper is got from
the same plant and comes
from the ripe fruit and the
outer layers are removed and
the center is used. There are
the black pepper is got from
green fruit. Now the outer part
of the fruit has the sharp stony
resin, and have characteristic
of black pepper. The middle has
the fragrance hence the more
agreeable taste and odor and
of white pepper.

Ginger family = ginger, cucumis
cuminum, grain of paradise, galbanum.
In ginger and cucumber the
resin is taken.

Valerian is used due to essential
oils and resins, oil is a simple
hydrocarbon. Resin is very com-
plicated.

Sweet flag is the essence of
and resin, it resembles the ginger
family.

Prime families: - Turpentine which
gives oil of Turpentine and rosent
Common turpentine, B. in gumdy pitch
Oils of Jupiter and cedar - oils of
cude and tur. These two latter
are got by distillation

Oils have several sub. Resins are
rather complicated

White sandal wood = oil of sandal
wood.

Green to Cyan Turpentine,

B. in cyan lentine

Catechu = ~~green~~ tree in South America
Cofobia, found in a South American
tree found in Cofobia in the trunk
resin is the most important part
and is much more agreeable than
thoat itself.

Cofobia resin is amorphous and comes
from the resin there is also a crystal
line

Balsam of Tolu. and Peru

Balsam of Tolu got by L. and

The pen also comes out with
the and is repeatedly done.

Benzoin. similar - look to toluidine
pen. and further benzoin acid

$$\begin{array}{r} 21219 \\ \hline 23\frac{1}{9} \end{array} \quad 115 \quad 118$$

Materna Medica Mar. 16. 1882

Tannic acid group. These
correspond to the vegetable astringents.
All veg astringents are these astringents
prop to some form of tannic acid or its
comp. and all the tannic acids
are astringent.

Lepo and ^{synonyma} ~~synonyma~~ galls.

These trees resemble the scrub
oak of this country and grow along
the eastern end of the sea.

And the gall is drawn out by the
incision caused by the sting of a fly
who makes a wound in a limb and
deposits her egg. This causes an inci-
sion and you get a hard, nodulated
growth covered with bark and as
it ripens the maggot eats out
and leaves an opening.

Altho oak are troubled by these or simi-
lar insects even in our own forests
Even forest.

When the gall is compl formed
you have a little globe of a dull

greenish color. In the best
galls you see no hole. in
the older ones you see the
hole. Taste a stringy astringent
and puckering, and sweetish.
Aleppo galls are the best. - others
are lighter in color and have
a large hole.

Blue galls have twice the amt
of tannic acid 80 to 100

Also have some gallic acid
abt 30%. You also sometimes
get very narrow galls of the
same structure

Gampner and true Chinese
galls are caused by a differ-
ent insect and on a different
tree. They have abt the same
prop of tannic acid and gallic
acid as the blue galls.

But there are not offshoot
Gallic and tannic acid
out most of gall structure
of gall. These two contain

seldom used.

Tannin. In the broadest sense it includes a group of sub. which have all the same prop. they are all acid in sol. for water with solts of iron. ppt golden and db when and are all astringent. They are supposed to be a decomposition products of certain veg. sub.

Gallic tannic acid blue black with solts of iron.

But every plant has some peculiarity in the tannic acid which it produces.

Gallic tannic acid is the one commonly used and is made by depriving the powdered gallic with very little H_2O or exposing to the dew on a damp night and then moisten with ether and stand 24 hours and exposed - again moisten and exposed and the two fluids distilled.

But it is never pure.

is never pure. or crystalline

It is of a buff color. sol in ether in which it has been added

It ppt all alkalies more or less. Slightly decomposed by heat. and gives pyru gallic acid and meta gallic acid in case

Glycerate of tannin. 1 of tannin & 2 of gly. spontaneously. ferments. and vinifies. When it is given it is not worth while to give it with any other sol. as it will prob ppt it

Gallic acid. gallic mac white H₂O of mouth. liquid than any and they are colorless without the acid distilled. then it is in the form of crystals. Does

sol in alcohol. ether. Only offic prep is the glycerate and

For local application is best For internal gallic acid.

because tannin decomposes into

Ther - stomach dose - 2 - 1, gum
citracae and cremonia. or
Sino Citachu. granum, rubus.
spina. Sponginate, red rose, holme
teline.

Material Med. Mar 23. 1882.

Bitter tonic. No medicines or nearly none
have only one action but have more than
one. Nearly all drugs have
an action of their own.

Bitter tonics are widely distributed through
out the vegetable kingdom. They
are not narcotizing and are
followed by increased appetite. They
are divided into mineral and veg
etables. Vegetables act through the
stomach. Bitterness exists in all
parts of the stomach drug.

Very mucous active principles are strychnine
and buccia. In many cases

In many cases the bitter principles
have never been extracted as such
most of them are soluble in water
and in the fluids of the mouth

Gentian family are particularly bitter
This family ~~are~~ ^{is} ~~are~~ poisonous there
are no alkaloids in ~~them~~ ^{these}, and
the bitter principles are neutral

Cinchona. have a bitterness and
alkaloids quinine, etc.

Magnolia have bitter principles
Sage, root, and Sarsaparilla family
all have active bitter principles
(Bartonia is active principle of bitter)

All these increase digestion and
promote assimilation and
increase the appetite

If these results are not obtained
stop them, they are given before
eating, if it is done before eating
take it after

All bitter tonics are not in the
or at least to a certain extent,

Gentian is most volatile of all
the bitter tonics, except perhaps the
alkaloids. It has a milky juice
leaves are all herbs no shrubs or
trees. The roots are large fleshy

and devoid of woody tissue found
in Europe and America. They are
4-5 feet high. root one or two
feet long. the root is the impor-
tant part. It grows on the high
lands. not in low lands or in
marshy districts. Come from
Switzerland and Spain *Gentiana lutea*
Gentiana lutea. this is the principal
kind and is the official

American gentian has small fibrous
roots.

Gentian purple is the active and
peculiar principle of the gentian powder
It is a glycoside.

It should never be given with a
free mineral alkali as the tonic effect
is destroyed. There is no tannin in this
drug. It should not be given with the
tincture of chloride of iron

May give me before and the other after
meals.

S. Sabacia. quassia. part used is the
wood. and is yellowish color

It has no Tannic acid.

Columba. order many species
Columbin is the active one of the acetic
principles. columbic acid. Lactyl
Gold thread. These are the
important. gold threads is a
common domestic bitter tonic.
Commocaulis is also one of the
important bitter tonics.

Magnolia (Cinnodendron). Cinn
is used as tonic. (active principle = magnos
lin) Cinnodendron = (Cinnodine.)
Commocaulis

Grass.

Asterium

" ~~Grass~~

Estuque

kin 799

Cremora

Geranium

~~Spina~~

Cubus.

Gentian

Coleman

Livian

~~Commocaulis~~ ~~with~~

fresh of all of
these mentioned

Sub for recitation Mar 28

Galls.	N. O	6	Page.	Dose in grm
Tannic acid	Lentaceae		667	
Gallii	Cupressaceae		86	.5
Catechu	Leg -		45-	1.
Tinct of "			370	X
Kino	Leguminosae		1439	4.
Tinct "			799	X. 6.
Krameria	Polygalaceae		1449	6.
Tinct "			800	.5
Geranium	Geraniaceae		1450	6.
Spiraea	Rosaceae		679	2.
Rubus	"		1337	3
Gentian	Gentianaceae		1235	
Ext "			677	1.
Tinct "			379	.3
Colombo			1446	6.
Tinct "			326	1
Guaiacum	Simarubaceae		1436	6.
Ext "			1194	40.
Tinct "			590	1.
Chamomile	Anthemideae		1456	2
Anthemis	"		420	4.
Oil			181	3
			984	.3

78

Infusa. catechu	763	50.
" " " Comp	763	4.
Anthemedi infusa	760	64.
Infusa. columbae	761	64.
" gentianae	765	32.
" gossypii	768	64.
Suppositories of tannic acid	1382	
Scrupum. rubi	1400	6.
1/2 " Krameriae Krameriae	1395	3.
Fluid extract of blackberry	609	6.
" " " columbae	597	4.
" " " gentianae	603	2.
" " " gossypii	603	4.
" " " rhodanthi	686	2.
Glycerinum gallici	685	
" tannici	686	2.
Troches of catechu	1474	
" tannic acid	1473	
Unguenta " "	1483	
Unguentum gallicae	1486	

Golds

Kino

Geconium


Rubus.

Spica

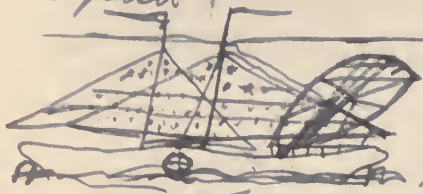
✓ Better Tuned are 3. input

 Columbus ant. mex.

 Russia

 Anthemisti

di Gentan



Tannic acid - soluble
part of glycine

Tannic acid should not be
prescribed with any other active principle
in medicinal chemical preparations.

Nearly all the antelope came under
their hands.

Dezolic acid is the latest chemical name for ~~carminic~~ ^{carminic} acid

Usual strength of dainties
is one to sixteen. (1-16)

Neither tannic or gallic acid showed
to given with the salt of iron

Gollic acid does not ppt the alkaloids
Tannic acid
Tannic gives ink with ac and some
salt of iron.
Gollic with some ink with one.
not with it

[redacted] Molena Med. Apr 13, 82
Opeloids

These are of varying comp
and prop. taste and solubility.
But they are all in sol. unite
with acids to form salts sol in
alcohol and ether most in H₂O.
apt to be crystalline and a
pretty def chem comp.

They were discovered in
1616 by a Dutch apothecary.

They are found in both
hemispheres. And as a rule in warm
regions.

As a rule the classes of
drugs have opeloids of a more
or less similar character

There is no part of the plant which may be said to yield the alkaloid exclusively but some parts are more adapted for obtaining it. In the *chama cinchona* may be said to come from the bark.

Nux vomica comes from the seed of a plant.

Rhus toxicaria comes from the leaves - and in all plants the old are prop in solution in the juice of the plant.

In the case of the poppy the juice is taken and dried and gave opium. As a rule the older the plant the more the quant of the alkaloid. Beyond a certain age however the diminish.


In temp parts of plants (leaves) it is greatest at maturity. This is true of fruit, but it should be taken just before it ripens and after it has attained its full size.

They seem to be degenerative products
of the plant waste products
as to speak

They have no particular
odor as a rule

The effect of cultivation on the
alk values: particularly opium and
cinchona. There are much more
increased greatly by cultivation

Some of these are very perishable others
are very liable to decay.
but as a rule they may be said
to be permanent.

 They are usually combined with
some vegetable acid forming a salt
Morphine with meconic acid
Quinine with a tannic acid
Rarely they are free as narcotine
Nuxvomica contains strychnine and hyoscyamine
They are extracted by replacing the
natural acid by a stronger acid
in this country H_2SO_4 is used
it is then extracted by means
of percolation. The stuff is next

doubtfull. all sol in alcohol
most of them are sol & ether &
chloroform. usually most in the
but if ~~it~~ But use a little more
sol in acid sol. These preparations
vary from in plants for $\frac{1}{1000}$ to $\frac{1}{10}$
Administration

The permanent ones may
be given as powder, or pill
form. ground up with sugar
or milk. In some cases they
may be made into pills as in
the case with quinine

If sol in water may be given
that. or if sol in acid sol
may be given in the manner
(mucil)

Lead up. *Nuxvomica*, *strychnine*
Buccia, *ignatia*, *gelsemium*, *gelsemium*

Materna Medica. April. 20. 1882

Order ~~Papav.~~ & Poppy family. They have
large showy flowers and secrete a
thick fluid. The fruit is in a capsule
and has numerous small seed and
on bruising yield a nauseous and
dull odor. The juice is sometimes irri-
tating to the skin. The plants are not
tropical as a rule. but is found
in temperate regions.

The leaves do not appear as early as the
horseradish. The leaves are broader than long
and have one or two lobes on the sides.

Sanguinaria with alk of food with
a sanguinaria. This contains several
colony matter dose of drug 1-4 gram
Tincture 1-8. dose of 60-80 drops.

Neither are much used.

Syrup of Sanguinaria dose 2-8 gram

This is rather useful in cough with
Poppy seeds is an annual growing as a weed
in meadow than the regular poppy.
has red petals from which is made
a bright red tincture

Papaver somniferum is the opium poppy. This is cultivated in our gardens. There are three kinds of this we find along the Mediterranean and two other kinds. *Papaver album* this is found in England and is cultivated in Persia and is used for production of Persian opium. The capsules are used as an article of Med separate from opium. Sometimes *papaverone* have been found in the capsules of the white poppy.

Opium is the dried juice of these plants and is obtained from the capsule. For the purpose of getting opium the poppies are cultivated as they produce larger amounts of the sub.

It comes to us of a rather dark color soft and comes mainly from Asia Minor where the cultivation is extensive. Sol in H_2O to extent 51-55% and yield 55 or 60% of mucous matter as an ash.

Egyptian opium. little yellow flakes Persian. in hard round balls.

European opium. seldom found - East Indian

opium goes to China and is used at home
Read up opium consumption.

Materials Med April 27, 1882

Among opium is the one used in the dregs
of medicine and some form. Acid Nitrate
Opium has 12 alkaloids, Morphine, Codeine,
Narcotine, Narceine, Thar the others of the
four most important ones.

Morphine - known to be a constituent
in the dregs of opium, 22% of the
found in the dregs of opium, the highest in
record. It is a white crystalline solid
to 1.5 gms. It is soluble in water, alcohol
1:1000, ether 1:1000, chloroform 1:1000, benzene
1:1000, carbon disulfide 1:1000, and in the
presence of water it is soluble in the way
to 1:1000. It is the most common
in the dregs of opium, 12% of the dregs.
It is a white crystalline solid.
Narcotine is a white crystalline solid.
Narceine is a white crystalline solid.
Thar is a white crystalline solid.

There is an official preparation of
Opium, Morphine, and Codeine.
Codeine is given in a variety of
forms and is used in the most life saving
doses to give relief in pain.

dose of morphine - 1 cent grain Co. dose
and Moraine 3-5 grains

Dose of opium (laudanum) 1-10
of alcohol. dose 20-40 drops
for an adult at time 25 to 30 drops
to the grain and dose 15-25 grains
Dose of tincture from the stramonium seed
and this dose sub removed dose
15-20 drops 6-8 grains

Complicated dose of opium, dose 15 grains
Dosed tincture and seed and
Dose of opium 1-4 of a drop dose 1/2
Comp. dose of opium 4-1
Dose of opium 1-4 of a drop dose 1/2
Dose of opium 1-4 of a drop dose 1/2
Dose of opium 1-4 of a drop dose 1/2
Dose of opium 1-4 of a drop dose 1/2
Dose of opium 1-4 of a drop dose 1/2

Dose of opium 1-4 of a drop dose 1/2
Dose of opium 1-4 of a drop dose 1/2
Dose of opium 1-4 of a drop dose 1/2
Dose of opium 1-4 of a drop dose 1/2
Dose of opium 1-4 of a drop dose 1/2
Dose of opium 1-4 of a drop dose 1/2

Dose of opium 1-4 of a drop dose 1/2
Dose of opium 1-4 of a drop dose 1/2
Dose of opium 1-4 of a drop dose 1/2
Dose of opium 1-4 of a drop dose 1/2
Dose of opium 1-4 of a drop dose 1/2
Dose of opium 1-4 of a drop dose 1/2

Butter is this is a great found in the
southern part of Europe and America
based partly in Germany. It is
atrophidia. root and leaves are both
offered. Atrophidia, the flowers and
hyocyanine, are the two alk. atroph.
a dried herb only as a sulphate.
This is a very good and useful herb
and root. The root is a little better
the root. hyocyanine, the alk.
are the same as the alk. but the
root is more of the same, an alk.
and from a 3rd hyocyanine.

Hyocyanine - D. alk.
All the group have the same alk.
of the same group.

Lobelia
Fruit of hy. f. l. & of hy. f. l. & of
The young leaves are the same
which have the same alk. principle.

Subjects for recitation May 4, 1882

	Page.	
<i>Nux Vomica</i>	960	
<i>strychnia</i>	1352	
<i>Brucia</i>	961	Loganiaceae
<i>igassumia</i>	961	
<i>Ignatia</i>	754	
<i>gelsemium</i>	674	
<i>Opium</i> (Papaver 1056)	1030	
<i>Sanguinaria</i>	1253	Papaveraceae
<i>Dulcamara</i>	518	
<i>Capsicum</i>	347	Solanaceae
<i>Belladonna</i>	274	
<i>Hyoscyamia</i>	747	
<i>Stemmarium</i>	1356	
<i>Duboisia</i>	577	
<i>Tobacco</i>	1404	
Prep of opium		
Aer. tincture of opium	1454	.5
Ex. tract " "	587	.03
Tincture " "	1453	1.
Decodurized tinct " "	1455	1.
Camphorated " "	1454	1.
Opium pill	1100	.06
Compound soap pill	1102	1.3
" opium pills		


Tinct. of opium
 Ex. of opium
 Tinct. " "
 Decayed Tinct. . . .
 Compound " " "
 Opium pill
 Compound "
 Extract of belladonna. for root and leaves
 Tincture "
 Extract of hyoscyamus.
 Tincture "

Compound specac. - powder		1.
Extract of belladonna	571	.81
" " " Fl.	597 497	.3
Tincture " "	1434	1.
Tincture of capsicum	1438	6.
Extract of hyoscyamus.	583	3.
Fl. " "	605	5
Tinct " "	1448	4.
Extract of nuxvomica	586	1-2
Fl. " " "		
Tincture " " "	1452	1.
Morphia sulphac.	937	.01

Percolation. 593

Repercolation 595.

Materia Med. 82. May 11.

 Cinchona. (Rubiaceae). This includes some very useful drugs. Specaccharum, cinchona and Cinchona are the 3 principal drugs. Some alkaloids are also produced.

It is confined to the Andes Mts. of S. America. The temperature of the region is temperate, and sometimes get freezing, and there is much rain and fog. They grow on the eastern slopes. The best kinds grow abt. 6000 ft above the sea.

They are either shrubs or trees and the trees are from 30 - 80 ft. They are evergreen. Leaves are opposite and oblong, and rather narrow.

The genus has from 34 - 40 species. The Cinchona trees are becoming more and more scarce.

The bark of the roots and of the whole tree are good.

Flat and gull. bark the flaccid bark.

comes from the roots and large branches
and trunk the galled come from the
small branches.

Color of the bark varies from a brown to a
dark red: and it is very brittle.

The specific kinds of bark are the
yellow bark *Cinchona flava* from *Cincho-*
na calycoglyphata comes from the
southern districts. This is the best of
the barks. And it has little depression
which are called digital depressions.

Cinchona pallida this is entirely out
of the market. *Cinchona rubra*. These
are the three kinds.

1854 war this plant was introduced
into Java. These barks have four alkaloids
quina & quinidia. cinchonina and cinchonidia
quina: quinidia, cinchonidia, and cin-
chonina this is the order of their value
There are also two others not of any
great value which have just been
discovered in India. *Green quina*
and *cinchona* tannic acid are they
are the acids with which these alk.
are in comp with.

Matena. Medicin May. 18. 1882.

Cinchonidia yield crystalline salts
then dried. Linnia. Linnidia. Cinchonidia
yield the same. All of these are
composed of decomp. with certain acids
yielding an other class of comp.

Their effects & the effects of them
is all the same. Linnia
is most important next quinia. This
has been thought to be even
more active. The dose. Cinchona
Cinchonidia. The dose of these is
oft twice as large as that of
quinia. They are less soluble than
quinia. Kinic acid is the acid
with which they are most asso-
ciated. The coloring matter of the
bark is cinchonin. bark. (tincture)

Decretum Cinchonae Florac. Comp. (Linnia)
(Linnia. Secret cinch. rubre.
refusum and comp. tincture) 1st
is the most important compound made
from the yellow bark. 2d from
the red bark. The yellow is

supposed to contain the most quinine. Extract and fluid extracts of cinchona. Sulphate of these compounds are used more in this country. Acetate and murate are used abroad.

When ordering a solution of these salts add to the solution a much dil H₂SO₄ as you have grams of the cinchona. Thus if you order 4 grams to 100 of fluid you would add to it 4 grams of H₂SO₄. This amt will always diss the cinchona.

Citrate of iron and quinine. Kelly of quinine. Valerate of quinine these are these valuable officinal compounds of quinine.

Syrup of Quinine with a valuable vehicle for giving quinine in

~~the~~ *Ipacuanthus glaucantherus* this is a valuable plant of the same order vitaceae subgenus and they grow in clumps like succulent bushes.

The bark is rough and tough and quite adgesy.

There is one alkaloid, is of a light
brown color. forming salts with the
acids. Called *Uricin*

Wine. Syrup, Fed & sweet powder.
This latter is the best for an emetic
Dose of powder 1 gm. fed ex. 1-1.5 gm.
Syrup 30 gm. as an expectorant
Syrup and wine 4 gm.

Coffee. This is a tree found in
Barbadoes, Africa, and East India. and
has been cultivated in Arabia

Mocha Java. Rio de Janeiro - the
order of these berries in order of their vol-
ume. Caffeine is the alkaloid

Tea. active principle is caffeine
Paulownia Lobelia

Erythroxylon thea *Coca*. *Ilex paraguensis*
Barbadoes

Morandi. alk. salts caffeine
Amino comp powder. 1189

Comp. un. pill. 1097
Meyr. 36. Salt of Na. W. Fe. 18
Syrup a sufficient quantity.

Materia Medica. May 25 82.

Stegnum.

Stegnum. (olk is to *stegnum*) *Chloro-ban* is a common name.

Physorhiza, *Chloro-ban*, *Colubia*. The seeds of this are very small. The plant grows commonly in North America. Is not much used. & kept as an expectorant, and in cough mixtures. It is a very dangerous drug as it may not cause vomiting and acts as a poison. has no other name in kit pref.

Maculatum conium. Leaves and fruit are used no good. Conium in the alk. is a bitter liquor. The extract of the fruit is the only thing worth using except the alk which is hard to get.

Ergot. This is a fungus growing on the head of

many grasses. and in
them. In a certain stage the
blanched name is chlorocephalus.

Found around the seed
and drary... it is usually taken
from rest.. It has none of the
rays in it. but is a separate
growth. it is propagated in two ways
by budding and by true fuctiona
tion. It sports very quickly and
should not be used when over
a year old.

Ex tract of Ergot. called ergotine
and there is also an alk-ergotine
& the ergotamine. has been found
to have active principles.

Chlorotic acid - obtained from the
drug. Its Extract is the best. prep
in the market at present.

Reprint Nature of Southern Europe
There are several kinds.

The roots look much like horse
radish roots. Active principles

in *aconitina*

European yielded most *aconitina*. Japanese
yield most *aconitina*. Both
yielding the two but in reverse proportion

Recitation. *Madame* 1st 1882

<i>Cinchona</i>	414
" <i>rubra</i>	415
" <i>flora</i>	Rubia 414
" <i>foliida</i>	414
<i>Quina</i>	424
<i>Quinidia</i>	424
<i>Cinchona</i>	424
<i>Cinchonidia</i>	424
<i>Ext. Cinchonae. flvae</i>	1440
<i>Secretum</i>	498
<i>Comp. Ext.</i>	1440
<i>Ext.</i> <i>rubra</i>	
<i>Comp.</i>	
<i>Secretum</i>	498
<i>Ext. & flvd. Ext. & .. rubra</i>	575
	<i>flvae</i>
<i>Citrate of quina</i>	
" <i>anderson</i>	628
<i>Pull</i>	1101
<i>Valerato</i>	1212

<i>Hypericum</i> rich.		783
Alk. <i>Emetia</i> <i>Emetia</i>		
Wine		1524
Syrup.		1395
Alk. Ex. tinct	Rubiacae	605
Powder		1118
Coffee.		210
Tea.		=
<i>Jobmanni</i>	Rutaceae	1084
Alk. <i>pulcarpin</i>		1084
Glex.	Aquifoliaceae	754
<i>Phyostigma</i>	Leguminosae	1078
Alk. <i>Phyostigma</i>		
Ergot	<i>Junci</i>	541
old <i>ergot</i>		543
Ex. tinct of Ergot		601
Powder.		
<i>Aconite</i>	Ranunculaceae	93
Alk. <i>Aconite</i>		98
<i>Conia</i>		
<i>Lobelia</i> .		

Materna Medica. May 1911. R.
~~Rectification~~ ~~XXXX~~ ~~XX~~

~~Rf Hydrarg. Biniodid gr $\frac{11}{11}$~~
~~Pulv. Iodid 3 $\frac{11}{11}$~~
~~Lymph. Aurantii~~
~~Aqual. a a 3 $\frac{11}{11}$~~

~~My~~
~~3 $\frac{11}{11}$ 3 I.D.~~

R Hydrarg. gr Biniodid gr iij
 Pulv. Iodid 3 iij ss
 Lymph. Aurantii 3 iv
 Ligna one teaspoonfull in
 water morning and evening

R Hyd. Biniodid gr xxx
 Pulv. opii "
 Et gentiana M
 Ligna one pill morning and
 evening after meals.

Materna. med. June 8, 1882

Verotum viride This ~~X~~ a native of
S. America alk. & *Verotum*
Verotum album is another kind

The emetics are the same - but
their action is very diff. They both
are - depurants and purgatives but
the album is much more active

Veridia ventidua ventidua are the
the other alk.

Choc Cochicum the crown is the part
used. *Cochichin* is the alk.

Digitalis (fox glove) *Bennet*. stem
comes the 2d year. long narrow leaves



The leaf is used -
and contains the active
principle. It is *digitalin*
digitaline is a glycoside. Two kinds
of powder. *fld* & *root* are the
parts used: *Imixture* is the best prep

Squillo. This is the size of a
large Bermuda onion and has the
general appearance of the onion - ex-
cept that the leaves are flat in-
stead of round

There has a good deal of gum.
an resin. Tanager and. syrup
of squill are two very fine prep.
of Sassailla. It had a great up.
at first a cure for Cephalic
Pain on the green fruit and rose
long with. and are a little thick
the middle of the ends
than in the middle.

and are quite long. 4-12 ft.

Merly and non merly. the
nearly indicates the kind of starch
Detection. Time, Comp. syrup. This
last is good for a flooring mix.
White sandal wood produces oil
of sandal wood
Cuso. oil cuisine. 13. blue spt of
hot tea. drink. good remedy for
the worm.

Pomgranate oil:—

Camellia

Pumpkin seed. are all good for
the worm.

Santonin.

Peppermint Mule fern

Cantharide of *actuo* *pini* n.
Cantharidin. times good for
 skin burnment. used also for
 blisters

Cleome cantharides.

Cuscuta

<i>Lenna</i>	<i>cantharus acid</i>	Leg	1287
<i>Clyanth</i>	<i>colocynthis</i>		445
<i>Eleterum</i>	<i>eleterii Cucurbitaceae</i>		520
<i>Filip</i>	<i>convolvuli</i> (resin)		796
morning glory family a dose 2 cammery			
<i>Scammony</i>	<i>filipini Convolvulaceae</i>		275
<i>Mandurke</i>	<i>Podophyllum</i>	used	276, 1123
<i>Gamboge</i>	<i>gamboge</i>		669
<i>Alor</i>	<i>plocin</i>		135
<i>Schutark</i>	<i>notarbaria acid</i>		1227

Barbendroeeae.
Guttiferae
Sibthorneae Sibthorneae.
Polemoniaceae

Cause.

Lapse Running jumping over
& other. Skating sitting down on a
cold stone or cold floor or any
other extraneous accident which
may produce inflammation of the joint.
There are the cause mentioned
by Sayre - in giving rise to inflam-
mation of the joint. He further says that
if this occurs - a person who formed
a strong constitution will
should it will go on unless
arrested. will go on to produce
deformity, ankylosis, and stunting
and may terminate in death.

He qualifies his remarks
by saying that the opinion
of Surgeons for the last 25
years and has been generally
taught and received. and
that age and authority have
rendered a sort of surgical
gospel. to which against which
would be heresy and blasphemy.

From the rather sarcastic tone
of these remarks it is quite evident
that Dr. Sayre does not agree
with the doctrine which has
been taught and received for so
long a time. He further says
also makes the statement that
most practitioners agree that it
cannot be cured by an anti-
scrophulous method of treatment.

A statement which is not worth
much - as much as the writer
does not mention the stage of the
disease. It is quite evident that
no system of treatment will
refuse destroyed bone. By restore
I mean restore it to its former
condition of course. as good an
ingrown outgrowth of osseous
tissue may take place.

But until we payed more
attention to local treatment and
we meet with much more success
in the treatment of this affection.

and argues that the fact alone
proves that the disease cannot
be constitutional for her cured
local treatment, influence a
constitutional disease, and
moreover he says that he has
been able to turn the majority
of her cases to a direct trans-
mission again and that most
of her patients are of ex-
cellent and vigorous con-
stitution, previous to her attack
and that the constitutional
disturbance came afterwards
and was entirely dependent
upon the hip trouble. Instead
of the opposite state of affairs.
A most modern British
State and that the common
error - that the effects of the
disease have been mistaken
for the consequences - and
for there are no doubt troubles
which are capable of producing

more severe constitutional trouble
than disease of the joint, and
we often see cases of phlebotomy
added - the cause of a following
upon such disease;

Dr Sayre gives a list of
143 Cases - which. 131 were:-
clearly traced to traumatic injury
directly over the joint, and about
- only 3 cases the disease he
traced to no definite cause
unless we consider a constitu-
tional cause a direct one.

From so. He later makes a
remark which seems a little
like a loop hole - to be used
- case of need - when he
says that in many of the
cases it requires the most careful
examination to find out the
cause - for the disease does
not usually follow directly
the injury, but that it sometimes
does not happen for months

after the accident so that the
patient and his friends have
naturally forgotten all about the
accident and its connection
with the disease - until especially
reminded of it by the investigation.

Looking at the matter in a
perfectly unprejudiced manner
I seem to me a thing

Says by the late remark.
hold on to speak - the bridge
behind him. It seems to me
that - any person suffering
from the matter what we need
get a history of a fall in
fact - from someone known
within a sphere of mother
and since it - a well known
fact that people are perfectly
willing to ascribe a cause
to ~~see~~ jump. Hence we
on the cause of every great
trouble. It seems a thing it
would be a pretty easy

matter to get such a history particularly
if the surgeon be looking for
such. Moreover he says with
my wife & family history
of her so called white patches
A person at times of a phlegmatic
family may present all the
appearance of white teeth
and often be misled they are
stricken down with the disease.

Says. Symp.

He divides the disease into three
diff stages. 1st Characterised by
inflammation. 2d. by Effusion.
obstruction and apparent elongation
3d by rupture of capsule & formation
in addition and Effusion
and shortening.

1st Stage detail In cases where
the information is of a corn
gravel. Such a case usually found
in those of a strumous habit
the Symp are obscure and diff
of diagnosis.

and are often referred to the knee
instead of the hip. Then, frequently
after better action the knees
in a supine position and
moving a child - treatise of
thouls - the knee when it is
really - the hip joint. The child
holder with a "calf" which when
once seen will not be forgotten
or mistaken for any thing else.

The child stands upon the heel
leg and throws the knee on
forward - the flexion of the knee
leg is also very imperfect. The
knee, on the affected side not
being capable of being flexed
sufficiently to cross it against
the knee when the other is resting
on it. Pressure upon the soft parts
over the joint - a direct pressure
over the knee - a - shut any
form of pressure which tends
to force the head of the femur
the joint produces pain

Rolling L limb slightly so as to be
here pressure. at once release
pain and if careful examination
be made with pressure and ex-
tension we differently will be find
a rule - diagnosis. We also get
- most cases an ankylosis of the
muscles of L especially thigh which
- many cases - very slight and
indeed usually - - the first stage

The - one set of Symp however
which a very characteristic even
- the very early stages of the
disease. If we look at the
muscles of the thigh we shall find
many of the quite strongly con-
tracted. And after a very struggle
and on attempting to flex the
leg we shall get more or less
resistance or compunct by or
twisting a spontaneous action
as he flex. which - a very
important and characteristic
Symptom

2d Stage

Here we have stage of effusion
which if much - and will
produce a peculiar deformity

An apparent elongation of the
limb & flattening of the foot on
the affected side the umbilicus being
lower on that side than on the
sound side. You also see at this
stage an apparent ankylosis
which is produced by numerous
contractions, and the thigh - slightly
flexed - thigh flexed and
leg on thigh. This is of course due
to reflex action and serves the
purpose of keeping the joint still
and safe - an indication of
the treatment to be followed

The flexion. The limb is very
firm. Pectus. active femur.
are so firmly contracted that the
whole limb moves as if off from
skeleton. That this does not
depend on any ankylosis at

the time - forced & puncturing the
joint. when for union is obtained
by it - by cutting the muscles.

Then the power of ligament enclosed
- a joint being open. & produce
false ankylosis. - Chondro

And if pneumonia here has made
by injecting the deep joint through
the pulse line. by which procedure
are. - Erysipelas. & Adenitis and
immobility of joint here produced.

3^d Stage. The disease if not
arrested. - Given rise to cancer and
perforation of the acetabulum or ulcer
ation and rupture of the capsule of
the joint. which causes an effusion
of a fluid which would migrate
the other synovial or peri. into
the surrounding tissue which eventually
produces - an opening some.
where - to the thigh & many
metastases at some distance
from the joint. This rupture often
given rise to the fact the cancer

That dislocation has taken place
for the second time of the occurrence
When the capsule ruptures and
no adhesions have taken place
the limb seems shorter and is adducted
and inverted and is almost
opposite to that of the 1st stage

This adhesion is spoken of
by Sir Charles Bell and doubtless
gives relief by removing to a certain
extent the head of the bone from
the inflamed cavity but the result
is a false dislocation - many
cases. Sayre says that no case
has true luxation been found in
thighs - some might be apparently
the same thing due to the position
of the head of femur - the
capsule separated from the bone
and being left in the joint
Then dislocating the upper part
of the tibia upwards and
somewhat the true luxation

In making a diagnosis hip disease
must not be confounded with
Sacro iliac disease. In the latter
we get the elongation of the limb
which we get in 1-2nd stage of hip
but hip elongation does not
appear in hip joint and is de-
tected by measurement of sup.
and spine to the malleolus.
There is - Sacro iliac then - no
elongation but there is pain. The
seat of pain - also chaff. - 2 sacro
iliac tender when press along
joint of sacrum and ilium. At
hip we also get a mobility of
femur joint and L movement -
Spine a sign of putting hand
under patient and lifting by
the spine which before disease &
curve will come down hand against
the table; Treatment of first stage
All authors whose opinions are
worth anything agree that the treat-
ment should be perfect rest

and mobility of the joint and
relief of the synovial membrane
from all irritating pressure from
together squeezed with tension
until removed and general stim-
ulation can may be seen. In such
case. The occasional application
of iodine or some other caustic
instant at the spot may give
some relief. Now when the joint
is left to itself we get a constant
and irritating pressure due to
the reflex contraction of the muscles
which irritating pressure not
only tends to irritate the joints
of pressure - but & the joint causes
and tends to exhaust the strength
and of child and bear one so
to speak the nervous resistance

Causing spasms during which
there is much suffering ac-
cording to the constant pressure of the two
inflamed surfaces there - more
or less - also of the head of

the line and of the acetabulum. Hence
we have to apply some method
of extension to remove the
constant and depressing cause
of suffering. We at this stage
not only wish to relieve pressure
to a great extent but also to keep
the joint quiet to do the time
on many kinds of apparatus.
But at the stage when we have
a chance of getting a cure without
loss of motion it is imperative that
the splint be removed or short
wholesome and passive motion of
a mild and gentle character. Great
care being taken to make gentle
extension to remove the inflamed
surface from contact & rub of
joint - and but we start up the
inflammation again. If the passive
motion is not given Sayre says
case of motion - the joint
resists. 2^d Stage laid over -
Treatment - similar to 1st & 2^d

Certain extent. We keep up the
^{respiration} pressure constantly. and in order
to get rid of the effusion we may
apply counter irritant - leeches over
joint. But best of all - to treat
the condition of patient and if
sufficient well to be able to move about
put on some kind of splint and
with high shoe and crutches get
him out - open air. If pressure
from effusion become very great and
there is danger of rupture of capsule
I may recommend puncture
I should great trephane and
evacuation of fluid. But may
cause the inflammation - if so severe a
character that the patient cannot
stand the slightest move and
has to be kept absolutely at rest

We have the stop secondary
to I suppose a rupture of the articular
capsule - a perforation of the articular
capsule being the more common
and the fluid within pure or purulent

Toxicology

Prof Wood

Mar 31. 1882.

[Jan H] Tartaric acid The reaction that of oxalic acid is nearly all its symptoms, but the blood in the stomach is fluid and remains in color. Chemical analysis is very diff. because it is used so commonly in household food. it is converted into carbonic by digestion and hence cannot be detected after absorption.

Evap to dry. extract with alcohol. drive off alcohol. adding a little H₂O in the

divided into 2 parts. 1st treat with potassium carb. and then add the other half and you get the acid tartrate of K or cream of tartar

which occurs as prismatic crystals of acid tartrate of K. This test is good for ammonium salts. tartrate of calcium potassium potassium carb. see specific tests

I'm trying the chroming test as not used. platinum foil

Treatment - to give soluble alk.

Alk. ~~potassium~~ ~~not~~. There are similarities in the common acid reactions of vomitin. but reaction with alk readily on -

The distinction of them is more extensive parts. Caustic potassium

Became the acids form comp with alk which are insol

Alk carb. & hydrot. and ammoniac comp are the principal ones.

~~NH₄H₂O~~ $\text{NH}_4\text{H}_2\text{O}$ affects the air strongly particularly. Treatment should be acids vomegar. The pain is greater than that of the fixed alkalis. The organs should be sealed up at once, as carbonates are formed by absorp. of carbonic acid from the air

Sodic hydrotic comp is a white powder.

Test for potassium hydrotate in the reverse of the tartaric test

Re cover in very rare in old foraging
Phosphorus. This is a very mis-
con- person. and a very common
one. The eyes do not come on very
quickly. And is apt to be over-look-
ed. Vermifuge and matches
are the principal sources of
~~most~~ ^{most} ~~poison~~ accidental poisoning.
Stick phosphorus is a poisonous
the Red is not and is ^{amorphous} crystalline
where as the stick is not.

Red phosphorus in matches is
not poison.

When used chemically the ends
are cooked in coffee or tea. If these
are cooked too long the phosphorus is oxidized
into hypophosphoric acid, or not
long enough, or he may not stir
it up.

A good many home elixirs before
eyes come on. you can get the
sticks to taste like garlic and
if seen in dark for a phosphorescent
glow.

Time 5-10 hours.

Common form. Nervous form.
Hemorrhagic form. These may be
single or combined.

Common form time 5-6 hours.
Face & throat swollen tongue diff
in deg. vomiting not bloody & dark
in phosphorescent for a slight greenish
odor. Colic, drunken. Urine older
men last for 24-36 hours then they
stop but the pulse is still feeble
and the patient collapsed.

But in for 2-4 days - sudden death
may take place - often between the
2 & 4 day. jaundice appears.

Some time of NH_3 in regard to jaundice
(Ar, St, Bu, Pto, NH_3 all then act much
alike and produce fatty deg.)

When the jaundice appears you
get headache - fatty matter in urine
(improb. def) delirium coma and death in for 6-12
days.

2d form Nervous.

It is dependent on action on nervous

system from membrane. cramp-
tongue. Jannochi was come - the
delirium came & death 6-12 days
3d - death - not so early. sometimes
bloody. stools bloody. life may be
prolonged until the cerebral tissue
may be diseased. Colic at last
and 3 or 3-4 week death from hem-
orrhage. which comes from all the
mucous even. and there are hemorrh-
agic spots under the skin.

Recovery from phos. is rare and
the prognosis is always very doubtful
unfavorable.

Phys. action of phos.

- 1 That it is oxidized at expense of
O in the tissue and blood
- 2 That it is absorbed in such and
acts after absorption
- 3 That phosphuretted hydrogen is
produced and causes death

The 1st. reaction is in the
arteries external - as in the rectum.

If it be injected free into the blood you
get an entire diff ret of symptoms

In regard to the 3d many say that
it is unsafe. ~~others say the same~~
And that the gum is all once oxidized
and become brownish. but the def
ends on the way in which it is
prepared. if prepared with nascent
hydrogen you get the regular color
and just as good appearance.
and Wood says that the last is
the ~~best~~ view.

Shorter period of death in 3 hours.
fats done for 1-3 grams but
a much smaller dose - a few
grains in children. Two matches -
a child 2 months old.

Two matches have proved fatal - one
if in adult. Match = ~~about~~ .01 g grams
if take - hits great arteries - first
- alimentary canal. if take 2 fine
ports or after cooking you get no
nutrition

Usually you get ecchymoses. - the
mucous membranes.

Fatty deg of liver first seen

and muscular tissue in the
principal region of the gastric
and intestinal gland are fatty degen-
erated. Liver may be enlarged
small. according to the time the
disease has lasted.

Inflammatory lesions are rare and
common with an - that have the
look of the large white kind of
New Bright.

Free phosphorus must be detected
in an analogous manner. There
is so much in the tissues of the
body - in fact in all fluids and
tissues it is found.

Autopsy. Have worn dark
and look for phosphorus in mouth and
throat then shake up the contents
of stomach and you will get
the luminosity which shows free
phosphorus.

It is oxidized very soon to phos-
phorus and then to phosphoric
acid.

1
O doc luminosity and red hot tail
which reduces NH_3 to nitrous oxide.
This is also applicable to phosphorized
hydrogen.

This luminosity is prevented by tin
pentac. NH_3 fumes. This latter is
important as NH_3 is lib. for tin

See. in the way you do the
analysis. as the cap. passes through
the condenser. at the point where
the vapour condenses. there you get
the luminosity. This is very delicate
1 match will give the answer for
1/2 hour. in 500 cc of H_2O

As a rule only 3/4 of the phosphor
passes over from the cell.

Treatment.

All fatty food must be stopped
Control it must not be used for
the reason that phosphor may or may
not be more easily absorbed.

Temperature old. (new of no good)
which has been exposed to air
and has gone.

and a supposed case of oxidation. if the poison it should be given in case doubt.

Medicines Chem. April 13. 82.

Lectures

Mentor symp. actions of the attendants
& Take a note of all ~~that are~~ who are present.

What is compound poisoning?
Where a person has been given and a drug has been given to cover the symptoms. This is compound poisoning. Instance strychnia then give morphia

Intoxication may hinder the onset of a poison.

Morphia and strychnia are antagonistic
The chemist should be given a hint as to what the poison is. as it saves time and material. This latter is a very important thing as the amt of material is always limited
He should make the ordinary preliminary test of the sub

and by so doing he may get an
important hint. But of more
important thing to do is to make
a microscopic examination of
the intestinal tract to discover
traces of the poison either. Crystals
or leaves. Digitalis has on it
two kinds of hair one pointed and
the other not pointed and very
bright. ~~very~~ very few other leaves
have this appearance. The color of
the contents is important as giving
a hint the odor: is one of the volatile
poisons. Fatty degeneration of liver &
heart indicates phos the staining
Grosser's sub may also cause
the appearance in a less degree
Organs should be divided into
more than two parts.

Begin careful examination of the
food for can tell what was last
eaten and also how long digestion
has been going on.

and thus get an idea of when
the food was taken.

Organs should always be
opened, by the ~~the~~ Medical Hammer
and not left for the chemist

Doct^r should always be on the
look out for pretended poisoning, they
may simulate the most of the symp^{ts}
but not all of them.

I. Lead with only mitiv^e which
comes under the ~~disfructants~~

Opium makes a class by itself the narcotics

Common have ~~hardly~~ a local action

I Take chap XI. XII.

~~Medicor~~

Toxicology. April 14. 1882

Wood wants to see me. Monday. in his
Laboratory.

~~Aspirin~~

Toxicology. April. 14. 1882.

Arsenic poisoning. In getting the
sulfimate of ac. heat the tube where
you think the sol. is to form and
you get large crystals.

Solvent of ac. poisoning, the numerous
(Arsenious oxide) white ac. is the most
common. A perfect pure oxide or
~~or~~ sulphide is not poisonous for
it is made in the ^(of elementary carbon) process.

If pure to moisture a part is
converted into the white ^(arsenious oxide) and is
poisonous. Metallic ac. (pure) is
not poisonous. This by exposure
undergoes a similar change
to the above and forms arsenious oxide.
(This metallic mixed with white
make the fly powder of the street)

Arsenic oxide (pentoxide) has
not given any cases of poison.
It is readily taken up by moisture
and becomes ^{converted} into a
moist mass.

White ac. = 2 kinds transparent

and change: The spore is more
sol in H₂O than the other in crystals.

Colors are very common colors
of poison. Ascorbate of copper.
I hold it green from green: one
to some thing and not a poison
no

Foulers sol in a poisonous sol
Donivans sol. 2 ~~poisons~~ made
of as and also hls Hg in 2 sol.

As ~~dist~~ in soft soap is a common source of poisoning and is used as an anti-parasitic.

Asenitized by hydrogen this is
formed a Winkler test for do
and is extremely poisonous.

Salts of arsenious an. is acids
and ^{poison} they have give few cases
of poisoning

11 Growth of Δ is used

Yellow sulphide as a color
has given rise to cases of
poisoning. particularly in children's
toys. where it has been used as
a coloring matter

Symp. after inges of white &c.
and the col of the very much
The many have 2 varieties of
acute. Acute = rapidly of symp
violence and rapid death.

This is not so common as the
sub acute = death in two weeks.
In cases of acute sym appear
in 1 hour after ingestion of
poison which is almost blue
white as at Pass green. white
gives no test except a sort
of burning sensation. Pass gives
a coppery taste.

Var. Sensation of heat great
thru^{out} ^{gastro} pain in the epigastrium
a little later a dysenteric pain
in the rectum.

Vomiting and purging may come
on rapidly. Very great prostration
fainting cold extremities & death
in from 4 to 5 hours.

Now there is another kind of
poisoning where the purging

pain and vomiting may be
absent but you have a great
deal of narcotism and the
patient dies in 6-8 hours.

The common form of the
acute sign comes on in 1-
1.5 hours. These may be delayed by
various causes and states of the
system. Opium may delay the
symptoms. Intoxication! One case
a man took 120 grains and signs
did not come on for 3 days -
death in 6.

When the signs do commence
you have old vomiting and
purging and prostration and
great old pain then comes
after a day or so a let up
of these symptoms. Thirst weak-
ness irregular heat cold exten-
sion. purpall deglutition (poss) rem
^{alive} this last for 200-400 hours
then comes a reaction and
we have the result of m^{over}flor
Old tender purpallandtic ^{over}REM

tenderness, sleeplessness, actin
of heart in affected face in
swollen an cyanotic. sometime
for 2-5th day you get an er
ruption due to actin^{on} in the
sweat glands in then effort
to eliminate. Specially you have
jaundice (and after death
you get fatty degeneration).

Intellect remains intact un
till a few hours before death
then convulsions and death
death in such a case is in for
2 days - 2 weeks. We may have
both in the acute and sub
acute bloody vomitus and
blood stools.

Death may not take place
untill after all we have
been eliminated death being
due to secondary conditions
due to inflammation but this
is rare.

Chronic form of poisoning

Usually due to accumulation
of Fowler's sol. in too large
dose. The poison (As) is not
an accumulative disease but
is rather one which is eliminated
and a chronic case it is taken
a little faster than eliminated.
Here you get olivaceous cyanosis
vomiting. Blood is then acid
rotary and have hemorrhages
from nose uterus throat &c.
face looks prematurely old
hyperaesthesia of skin. Trem-
bling of the limbs (paralysis)
death in from a ^{month} ~~week~~ to three
years.

Prisoning due to external
application. Such an canker
here we should always have an
eruption at the point of applica-
tion. Ordinarily in case extra
acute or so acute, in an average
death comes in from ~~1 day~~ ^{3 days}
Shortest period is 20 minutes

largest 2 years & a case of chemi.
2 or 3 grams of white or in an
fetus dose $\frac{1}{8}$ of a gram has
produced serious results
2 grams given & divided & de-
scribed as entering through 5 days
has proved fatal.
($\frac{1}{2}$ oz of Fowler = 2 grams of ac.)

The first is acute great inflam-
mation of mucous membrane of stomach &
intestine. In the chemi and
2nd acute you see no signs
of active inflammation. The only
thing is a pale swelling of the
gastro glands & also color due
to fatty deg. same in tube of
glands in inter parietal & solitary
glands.

Fatty degeneration comes in
in some cases. with remark-
able rapidity. beginning in 3-4
hours. complete in 10. at times
preservation of the tissue in-

a thing to be observed. as this is
quite characteristic of as - Prof
Wood mentions a case where the
body was preserved for 26 years.

Discolor spots may be found beneath
the skin. There are no lesions of
skin. mouth or esophagus. In
the stomach you get ulceration
at times, great redness and at
times perforation but this is not
common. The redness may
cover the whole surface or may
be confined to spots or may
only affect the fold of the ingu

When carefully ex we can usually
find traces of an as little white or
yellow patches. white & death has
come a quick yellow (sulfide of
as) if death has been delayed.

We find the same look in the upper
part of small intestine and in the
rectum & colon.

The solid ~~lesions~~ organs are
affected least kidney liver and

muscular tissue are as a rule
fatty degenerated. As may remain in the
tissue for a long time. Can be
on record where it was done
in 16 days. but when given in
larger doses, it is not so easily
eliminated.

In a case of chronic arsenic
poisoning 6 1/2 weeks after the
last dose.

Medical chem. recitation. April 20, 88

What are the common signs of irritant
poisoning. Coffee ground vomitus. Blood-pain
in stomach. In chronic you often get pain
etc. How frequent are cases of H_2SO_4 poisoning.

Give signs of H_2SO_4 poisoning. Pain in
abdomen; burning of mucous men of esophagus
Coffee ground vomitus. Acid reaction.
Constipation. Retention suppression of urine.

Bladder empty or filled with bloody fluid.

Dilute H_2SO_4 will brown H_2O when exposed
to the air until you get a green layer
and the red stain becomes brown.

What salt is often mixed with H_2SO_4
One Indigo. This is often mixed

with H_2SO_4 and its marked blue color readily leads to its detection.

Dilute H_2SO_4 does not cause such violent symp and you do not get perforations so often and rarely have perforation.

Toxic dose is 2 grams. Antidotes are chalk, lime, $MgCO_3$. Mg is preferable to the others, as it acts also as a cathartic and thus assists in its removal. Give tests for H_2SO_4 .

$BaCl_2$ does not distinguish between free and combined H_2SO_4 .

Diff betw between H_2SO_4 & HNO_3 .

Color of vomitus, is yellow. pain more diff and it also attacks the air passages.

Spots on clothing are dark yellow, as they are on the mucous mem. This yellow color is due to formation of nitrate of albumen. Nitric acid also gives out an odor due to nitrous fumes. H_2SO_4 is odorless.

Tests for nitro acids. = $FeSO_4 + H_2SO_4$

Symp of HCl resemble those of Nitric acid, as their solubility is abt the same. Both affect the air passages.

Detection is very hard.

M. If free HCl is present you need not add nitric acid for a crystal of nitre will do. the same is true in the reverse. where you may add a grain of common salt. As Cl. is sol in NH_4OH . and sodic hyposulphate, this test alone does not distinguish it from all of the hydrogen acids are ppt. by this (hydroiodic, hydrobromic, hydrocyanic)

Principal sources of oxalic poisoning
Salts of lemon. These may consist of two ingredients: Pure oxalic may be rendered
the term.

The crystals of oxalic acid and MgSO_4 cannot be distinguished, as for the look grey. Oxalic does not char.

Epsom salt sol in alcohol. insol in alcohol. sol in H_2O . Oxalic sol in H_2O sol in alcohol. In a tube you can by gentle heating sublime oxalic

Oxalic acid is very sour and on account of its extreme solubility it is almost impossible to give it cumm.

Symptoms: pain in stomach vomiting

if patient lives several hours you get purging
vomitus. may be bloody or not but if
they usually get blood. after blood you
get a weak pulse collapse and prostration
pains when given in dilute form.

Shortest fatal period is 3 minutes.

Chalk is best in this case because it is much
less sol. than Mg. next is lime H_2O

If acid & strong you use not the stom-
ach pump. if weak do other wise

Toke. Begin with post mortem look
possible and go to an poisoning

Toxicology. April. 26. 1882.

The elimination of ac is usually complete in 3 weeks. One case is on record where found after 6 weeks.

The ac is not distributed uniformly through the body. Some times taking more part in the liver which has more than any other tissue. here the max amt is greatest in about 15 hours. providing that absorption commenced at once ac may have acc amt of ac = to 6 grains of white ac. At this time fatty degeneration has commenced. The time at which it commences is uncertain it may commence in 7 hours. In one case it was much advanced in ten hours. in one case it had started well in 7 hours.

Some claim that the greatest amt ~~has~~ contained is stored up in the liver and nervous tissue. But they performed the test by war-ther test. And as some sulphur and organic matter

may be deposited the weight of the
till may be measured and hence
an error. Liver and Kid have abt
the same amt in proportion to their
weight. The amt in the stomach
and intestines depends upon the
amt of absorption.

Treatment of acute arsepoisoning. The great thing is to get rid of
the poisonous matter. Large amts
of ar may adhere to walls of stom-
ach. Use stomach pump if you
can get one if not, use grasy tho
Myrtol and water. Or dry thing
which will dissolve the ar and cause
vomiting at same time. Freshly prep
ferrous hydrate is the very best anti-
dote. Ingestion of alk comp are
very injurious. Because the ar comp
are so soluble. When the pump
is used stream should be passed
for an hour or so.

Magnesia is the next best thing
in order to remove that which

has been thrown into the intestine
give castor oil to carry it through quick-
ly. In case which go on favorably, the
urine should be tested daily for the
presence of ac.

Prop and Test for ac. Two former
trials and of course. 1st in an large
warmer and in much more sol
and in H₂O and is not common-
ed in the form which is often met
found in octahedral and is sided
crystals. in slightly sol in boiling
H₂O slightly in cold. Weight of ac.

A heaping tea spoonful weighs
150 grains. Table spoonfull 520 grains

Sh of white ac is much greater
than that of H₂O. but it flops in
in water to a great extent this dis-
it from other providers.

And this should always be done
at once it be made as should
the test for volatility as not many
are volatile. put a little on a
knife blade and it will form

an does $HgCl$. $HgCl_2$ & ~~Sol~~ ammon-
iac ~~Sal~~ ammoniac.

Melotte an is also sol in well in
the white oxide. Another very ready
test which can be applied to all the
compounds of ar. Heat with a
strong reducing agent (best, sodic flux
flux). This is better than the potassic because
the latter takes up the, which rather
hinders the test. This tube gives off an
odor of garlic, and you get a deposit
of white ar. corresponding to the amt of
ar in the tube. This is very valuable
in all legal cases.

Arsenite - yellow. arsenate

$CuSO_4$ + arsenic acid = light green arsenite
of Cu. $CuSO_4$ + arsenic acid = blue green
~~arsenate of Cu~~ arsenite. Both the Cu &
As sol are sol in Mineral acid and
 NH_4OH

The sulfuretted hydrogen test is
one of the most useful tests for it
can be applied to powder as well
as sol.

Common As_2O_3 may be converted into As_2O_5 by boiling with As_2O_5 NH_4O_3 or with free Cl .

Formation of triple arsenate of these crystals are similar to those of triple phosphate. The atom of phosphorus being replaced by an atom of As .
($\text{Mg NH}_4 \text{P}_3\text{O}_9$). ($\text{Mg NH}_4 \text{As}_3\text{O}_9$)

~~Murexide and Leuchs. Test are the two most important test of all.~~

Method (Arsenic acid test for any. Iron is oxidized with Cl or HNO_3 and is acidified with HCl and a vol of uranium added and the test is just the same as that for the phosphate. The water is next to none before). If you destroy the organic matter and pass H_2S the sulphurated hydrogen you get a ppt even though you have not metal there. This is due to organic sulphides.

No color dependence should be placed upon the color of the ppt.

defect obtained by passing H_2S .

With the aid of the white fused mass in water all the tests for arsenic acid may be obtained).

Remake test. acidulate the sol with HCl and boil the copper removes a part of arsenic from the sol. and it is deposited on the copper.

Wash and dry between filter paper and this gives a sublimate of copper. The test kind of a tube is one with a bulb and a stem neck.

Make test. Method of analyzing

Medical Chem. April 7. 1882.

What are the first visible appearances of
+125 poisoning. H_2 is useful in the
test for $HClO_4$ acid for it eliminates the a
large amt of organic matter.

Essential difference between acid and alk
poison. reaction of similar diff. but
in some alk reactions involving and
give a soapy feel. Virega is water soluble
for some of the more greenish yellow
as it is alk. Better than either of these
in milk or olive oil.

In case of ~~poisoning~~ poisoning
it is only case of poisoning with
any free water fluid
Most of the cells of the alk cells
with body and poison.

Ground glass will break & is muscle

Never saw any alk in a case of poisoning
but get some in a case of poisoning
but get some in a case of poisoning

Some of the H_2 is absorbed as such
as it is feared by an examination of
the entire

How do we know that phosphated hydrogen
absorbed ~~the~~ as a product of ^{the} reaction, there
a lot of nitrate of silver which is reduced
and as the bath is not common you
I know that there is no free phosphorus

Terpentine should be given in
large doses. for if given in small
doses it is absorbed and acts as a
poison. but given in large doses
it passes rapidly through the in-
testine and does not get absorbed.
Large doses of it is a temporary
antiseptic.

If the patient is
treated with terpentine the ph. presence
is obscured. Nitrate of silver & hydro-
gen also does this.

As and St. for Dr. C. T. Watson.

Toxicology. April 28, 1882.

Test for as in blue paper. Marsh's test is the only reliable one for papers of all the colors and for cloth.

Cut the paper into small pieces moisten with conc. H_2SO_4 and char on a H_2O bath to show that they are all made from H_2SO_4 .

St. Martin with NH_3 and you get an azure blue due to copper then add a drop of ac. NH_3 and you get a canary yellow. due to a minute of silver sol in H_2O of NH_3 and the mineral acids)

Moisten the charred paper with H_2O & filter or rather wash with water and filter. In performing Marsh's test care must be taken to add neither too much or too little, & always test your reagents in doing the test in the ordinary. In H_2SO_4 is sp. 2 contain As . And this is done by testing the flame with a porcelain plate & the mixture has been added. In criminal cases

you should also test the delivery tube
for 15 minutes to be sure that no iron
is formed in the tube. Haematuria is a
common symp associated with arsenic
retted hydrogen poisoning. and you also
have nausea. In a moist atmosphere
the as in wall paper evolves a little
sulphuretted hydrogen. This is inhaled as
to the dust, derived from the paper.
But when the paper is of a bright color
you can sometimes get a green dust,
before the arsenic and. Sometimes where
dusting is not common. Here we
get two sources, gas and dust.
And the fact that dust is in the
air has been proved by filling water
with cotton or cotton and passing a
current of air through bringing it into
contact with a solution of silver and from this
it has been obtained.

After the gas has been tested then in-
terference to sol. and wait for a minute
or two.

Sources of chronic poisoning most common
in wall paper. Poison green

all of the aniline pigments are apt
to contain Ar.

Shield green arsenite of copper has
80% of copper and in the olden times
was a most common source of poison-
ing. Ar. Opium & Urota are sources
of poisoning and as they are applied
to toys they give rise to many cases
of poisoning.

Another source is the use in the arts as
fly paper for instance, which is very poison-
ous. Coloring of clothing is also apt to
contain Ar and gives rise to many cases
of poisoning. If fabric has wool in it
dyeing the organic matter with HCl, felter
+ ClO₂, felter and wash and extract with
H₂O and mico does not fark.

Artificial flowers give rise to many
cases of poisoning - partic green and
red.

Water color paints pale light green
have Ar. The dark green is also
poison although not due to Ar.
Candy is often colored with Ar.

Porter the red. and green. Poiri
Green. is colored with Poiri green
Chrome yellow is colored used as a
coloring matter

Amaline pig used in confec are some
times poisonous.

It is colored is often colored with a
pig containing As.

Artemical groups containing blackberry
Jelly forms

To make the wine is detected very
easily as follows add to 50 cc
of wine. + 10 cc of sat. soln of Pb
nam and filter evap. + 10 drop of
H₂A + 10 cc of amyl alcohol and
shake. and after the separation the
alcohol is colorless if pure but
if not it is more or less red
if it contains some acid alcohol
will be yellow. if colored with litmus
set violet or red.

To test color put some in a
test tube + a little NH₃ and if it
becomes colorless it must be
poisonous

Wall paper. We cannot say from the
look of a paper whether it has ac or
not. We may even have the same
pattern and even one has ac. other
not. (Blue shield green = chrome yellow
+ prussian blue)
Paper used for wrapping candy. often
has a large amt of ac.
Glossed papers often have an immense
amt of ac. in them and are very danger-
ous. and are used by young children in
toys. Playing cards often have a large
amt of ac

Toxicology. May 5. 1852.

Symp of chemia as poisoning. Most of these
symp are due to the dust set free from
the paper. which comes in contact with
the mucous mem of the throat and
lungs. There is catarrh. cough and the
patient has apparently a bad cold.

There is also constipation. And
as a small amt is swallowed we
get dyspepsia and if the inflam gets
into the intestine, we get a little di-
arrhea. Later we have symp due to the
action of the ac on the nervous system
and you get what is known as
nervous prostration. And all the
symp of these obscure diseases. are may
be caused by ac. (Carpenter often
have a large amt. of ac.)

Women are much more frequently
affected than the men because they
are in the room more - Individual
peculiarity has much to do with
this. Death is often caused by
this form of poisoning

Epileptic convulsions may be caused
by the fume of ac. poisoning.
Antimony poisoning

Sources are two tartar emetic & the
fer chloride. The pure metal and the
sulphides are not poison.

Tartar emetic is a double tartaric of
antimony and potassium. $KSFOF$.

This has abt 44% of antimony.

It has caused many deaths
both from chronic and acute poisoning.

When taken in large doses it acts
as a violent emetic and thus you
get rid of it.

Dose of antimony has 32 gr to pint of
strong wine. Ointment 120 gr to the
ounce.

When a large dose is taken.
you get an acid tart. (which as has
not but as a whole it resembles ac
poisoning). Then vomiting and pur-
ging which may be absent. The
mouth and throat and the
nuculari green or covered with
an opthous crust.

Solivation is also seen (as it is
sometimes in ac). Vomitus often
bloody as are the stools, clamps in
the clammor, sometimes of same.

The symp strongly resemble those
of acute dysentery.

Antimony in first case progresses
rapidly and fatally toward death
then the dose is small it goes
on until recovery commences which
is also regular and progressive
In ac there are intermissions
and let ups which differ from the
symp of ac antimony which is
progressive and regular.

In an over the proportion is
in proportion to the amt given.
In cases of chronic poisoning there
is an increase in the amt of
urine, and the urine has large
amts. In ac the urine is dimi-
nished. Intestinal symp resen-
ble those of chronic dysentery.
External application of antimony

partur the chlorides which is found
on most cancer points, causes
cases of poisoning. When these in-
jected give often get a pustular
eruption on the surface of the body.

Port martin. examination.
Ophthalmic crust. fatty degeneration.
inflammatory process which may
be confined to duodenum, cecum,
rectum and ileum ileum. The greater
part of the small intestine being
skipped. Fetal dose. it hard
to determine. $\frac{1}{16}$ of a grain has
given. severe effects. But this
is & optimal. The smallest fetal
dose was $\frac{3}{4}$ gr in child & gr. for
an adult. death ordinarily takes
place in from 2 - 6 days.

Treatment. Is to form the vom-
it. Give tartaric or gallic acid.
These are always at hand and
are very valuable remedies.
Gallic completely paralyzes the
autonomic system. The is gallic

The dried (terrestrial) is very sol
in water.

K₂SO₄ in ppt from sol by nitric acid
one must be taken not to add too much
acid as it would be excess. of nitric
a. in an excess of H₂T then the detection
gives it from the similar B₁ compound.

By the addition the oxygen is fixed
Lemmer has given a black deposit
on the copper. in. It then is as and the
subliment is nearer the flame and the
crystals are vesiculate rather than octahedral.

Maunder test is of great value.
And if the delivery tube is heated the
Sb is deposited on both sides.

And if heated in a slow current
of H₂S. you get a yellow sulfide
of As and an orange sulfide of
Sb. If it is passed through
Sol of AgNO₃. in you get ppt of
white Ag. in case of Sb. you get a
mixture = SbAg.

As sol in some hypochlorite. It is
ammonium sulfide

In case of Pb poisoning all the urine which can be obtained should be. For often you can detect the Pb by concentrating and performing Lemche's test.

Pb can often be detected in the body long after death. And it preserves the tissue.

The case of those from the ten chloride are very few and its action is that of a emetic and in the cases which have been seen the intestines have been very much injured.

Medic. Chem. Toxicology. May 11, 82.
What is the treatment in a case of acute Arsenism. I Ans. Give emetic and then give a sol. of fresh ferrous hydroxide, which is made by ppt ferrous cl. with NH_4OH . And wash the filtrate thoroughly, until all NH_4 is removed. because any free NH_3 will be arsenite of NH_3 which is very sol. and is easily absorbed. a tox. of.

My with the - still better -

How test white As powder.?

Ans. It floats on water, and is a heavy white powder - volatilizes on heating, and gives off white fumes, which may be condensed on a slide. Gives a bluish tint in the lower part of the flame.

Sulphuretted hydrogen added to the sub gives a yellow color - Ranche test and the reduction tests with ~~zinc~~ carbonato. blood. flux which is made by squinting zinc carbonato.

Liquor. tests as those with Arsenite and with. Sulfate of copper, and with cerium. The last is used to get the quantitative test. The proportions of which are the same in test for phosphorus.

In the analysis the phosphorus acid is separated by the sulphuretted hydrogen and thus all part of the reaction being due to phosphoric acid is removed.

Means of distinguishing Pt. As and St. by means of the? Diff in color and solubility. and in volatility

In the mirror - with intense acid
and crop - and test with white of egg -
brown - of an. grey in St.

Has two a green. powder quickly of.
Ac P. Am. Heat it reduction test.

$NH_3H_2O + AgNO_3$ This - done & adding
the NH_3 & the powder get a blue due to
Cu - then add the Ag. and you get
an accretion of Ag. which is a canary color

In testing paper you marsten the paper
with the H_2SO_4 and a cham. filter and
add to acid in the H. generator

All the test mentioned on the
paper are watch glass. tests. only a
grain of the powder being used. with one
drop of the reagent

What are the principal sources of
ASTOT poison? One the taste
emetic and the triphloride and
the prep wine of antimony - and
the ornament of antimony

It is very rough wine and for
an. And the taste is acrid
and the crystals are tetrahedral

and laye. then points distinguish
it from an - What are the symp-
toms of St. poison - ?

What is the progress in St. ?
In chronic poisoning the symp resemble
those of chronic dysentery. for which
it is distinguished only by the
examination of the urine. This is
the only way you can distinguish
it from dysentery. The amt of
the urine is increased in chronic
due to stimulation of the kidney.

What are the post mortem appearances
? Ans. There is inflam of the diges-
tive tract and with exception of small
intestine which is mostly skipped. fatty
degen is also seen partly in the liver

What would you expect to see
of chronic poison ? Ans. fatty degen
and no emutation

Mucous men of air passages is more
apt to be inflamed in autum than
in an -

What do we see when St. has

Green. applied externally and it
is also green - at times when taken
internally. What is the antidote?
Ans. Any sol containing gellic
and tartaric acid.

Gellic acid can only be used as
a reagent when for quantitative work
when the sol is in form of Teicholud.

What are some of the ready tests
for As_2S_3 O.T. & I. Ans. Remer's test
Marsh's test. Bedside test is sol-
ubility. taste charming and gives odor of
egg. dipped into flame get. mass of
purple and blue. What are colors
of Sb and K . Sulfhydryl by
Hager & powder = orange color.
added to sol - plus HCl also
gives orange color.

Orange sulphide of Sb is sol in
 HCl and in this sol. if dropped
into H_2O get a white ppt of As_2S_3
chloride. sol in HCl and HNO_3 .

Take next two chapters. Hg, Cu,

Toxicology. May. 12. 1882

Hg.

Hg and all its comp which are sol in gastric juice are poison. The principal sources are $HgCl_2$, H_2Cl , & Hg .

Calomel prob gives rise to the most cases of poisoning because it is impure and has some of the soluble Cl . or it contains some which is sol in the gastric juices.

If you treat $HgCl$ with $HCl + H_2O$ and you get no mercury - but if you add to it gastric juice you do get it. Hence never give calomel during digestion.

Hg is the principal source of chronic poisoning due to the fact that Hg is vol at any temp above 40 deg F. And in the mines and laboratories where it is used you get many cases.

$HgCl_2$ gives many cases of poison from its extensive use for an antiparasitic. And in the taking of sol which were intended as external lotions. All the sol forms of Hg may be absorbed from any

surface: Hg is absorbed through the skin as such, and is not changed of ~~the~~ And has been found as such.

Symp of Hg Cl₂ are local and general. The local are corrosive and as much as acids and alk.

Local infln of the whole - alimentary canal. Mouth and oesophagus are often covered with a white coat due to an obliterative of Hg After exposure to air for any time recover harm.

Vomiting is rapid first contents of stomach then become often bloody more so than As or Sb due to the greater corrosive action. The diarrhoea is more apt to be bloody. The abd pain is intense. Vomiting comes quicker than in As. Expectations are frequent and bloody, urine is almost suppressed and very scanty and you have well-known signs of acute nephritis. This is due to effort of kid

to eliminate *Leptotheca*. Breath in-
fected and has peculiar odor and
in some cases we get *Solomon*
but not commonly. Urinary toward
end. Patient seems a little better
but the Leucic grows slower and
weaker. respiration diff. and derm
in from a few hours to a day or two
Nervous signs. Loss of sensation
in lower part of body, upper part
is normal. Apr eruption is also
common much resembling erythema
but last only 4 days.

Chronic poisoning. commonly
called professional poisoning from
the frequency with which
Mercury is used as effective.
This may be caused by frequent
doses of Hg. in medicine. The
first sign is that the gums are
a little swollen and tender.
This tenderness is prevented by keeping
the gums together. Next is the
diarrhea. If the action is to Hg

has been given a long time we
get a purple line due to the
decomp of food in the place of
min. H_2S which unites with Hg .
and gives sulphide of Hg . Same
things cause the rings in lead and
copper. poison

In long cases jaw and cheek
may become ulcerated, and this
may be mistaken for syphilis.
Rash in the lower also. comes
which closely resemble those of
syphilis. Potent in face and
depressive for action of Hg on the blood
fermenting. and epileptic convulsions
Tenderness of gums. fall in week
and describe as the warning
symptoms. The production of sal
ivation varies much with the dose
and the affected time of its appear
ance varies much. but teeth have
to come in show. It may appear
after having ceased. when the Hg
being given in the mean time of

Solvation may be produced by many
sub. digests. Dr. E. L. Drach of K
false teeth, B. compound. and all
uniting compounds applied to unit
will produce them.

Mercurial palsy and solvation
are very well seen together.

Signs of Hg applied external are
all the same. solvation and exup-
tion are more common.

Post Mortem. Evidences of inflammation
and unitation, the white coat, which
gives down as far as a pair of mice
intestines. When exposed to air-
stomach looks black. The tongue
is swollen and the papillae much
enlarged. Kidneys are acutely in-
flamed. (Acute parenchyma neph.)
In chronic you have (Chronic
interstit). Fatal dose: 3 gr
of corrosive sub. in the fatal
dose on an average. Time 1-5
days. In chronic duration of
life is uncertain.

Treatment for vomiting
Give all the voluemen as other in
the best antidote forming an
almost insoluble precipitate of
Hg. Eggs. milk are the two best
Alum may be used. The starch
may be mixed & washing the
stomach with water and stirring
through a cloth. But it may
be used. Food mixed

Use of stomach pump is doubtful.
For symptoms of chronic
Chloride of potassium should
be used. Test volatility. This
is true unless in combination
with fixed acid. There is no charring.
H₂S gives black color with HgCl₂ & HgCl
HgCl₂ if warmed in a tube gives
long beakulate crystals. If a little
soln be mixed with the crop
and you get a sublimate of mer
Hg. which may be condensed
into a large one by rubbing together
with a glass rod or tube.

Toxicology May 18. 1852.

What are the principal sources of Hg poisoning. Carbide sub furnishes the cause of acute poison. Hg. and calomel the cause of chronic.

What employments give cause of chronic poison. What is the best preventive. again Hg poison in factories. Am sprinkle the floor with NH_4HCO_3 This has been found to do a good deal to prevent this.

$\text{HgCl}_2 = \text{R}_2\text{C}$. HgCl ore.

HgCl is a heavy crystalline sub. very sol in H₂O. alcohol. ether and ~~chloroform~~. Calomel is often contaminated with corrosive sub. How are HgCl and HgCl_2 made?

What are the signs of a large dose of HgCl_2 .. Ans. White smoke. from throat. vomiting violent and bloody. bloody. nauseous taste. white coat. More apt to get blood because it is more corrosive.

Can we have acute poison from the external application of the poison.

Lead. and silver. cu. give
a line around the gums, as do
all mineral. poisons which. give a
black anephride. P. H. H. gives blue
line. Cu. more of a purple.

What is the time of death. Ans.
usually in twenty four days.

What is the antidote. Ans. oil
mutter. Eggs. milk. flour. There
four are ~~best~~. oil of H. G. which
in sol in an of each of the others
to much should not be given.

P. M. offenayer. Mouth and
respiration have a white color due
to oil of H. G. turns brown on ex-
posure to air. stomach and
all parts of the tract touched by
the poison. Degeneration some times
taken place: Kidney. in acute gives
acute parenchymatous nephritis.

Chronic. produces chronic and
interstitial

What are the signs of chronic
poison? Ans. tenderness of gums.

and teeth, diarrhoea - one of the
first symp. What is solution?
An increased secretion
of saliva. There may come on quite
rapidly. quickest time on record in 1 1/2 hours.
with quickest time on record - some
persons are much more easily
affected than adults, others, children
are not so easily affected. Salivates
but are easily affected by the poison.

How detect. Hg. comp. by ready
tests. P. Crystalline form. In case of
Hg Cl₂. Colomel. ammonia. Hg Cl₂.
flures and get white fumes. and condense
on a piece of warm glass. cold glass.
and are needle shape.

~~A. J. green greenish yellow~~
~~& Hood. H₂O. H₂SO₄ H₂PO₄~~

Sulphide of Hg. in sol only. aqua Regia
[C] Ph. Ca next time

Toxicology May. 19. 1882.

Hg is usually gotten rid of in a week or so. When a person has been treated by mercurials it may be found for a long time a year over.

Hg has been found in the urine due to inhalation of Hg fumes.

Lead poisoning

This is often met with particularly in the chimney furr. Most frequent form is the furring of water through lead pipes. If H₂O it took a lead pipe - much more than hard water. & water containing an excess of alk carbonate, attack it much more vigorously. Oxygen dissolved in the water unites with lead to form ox of Pb a common litharge. This meets the ox of carbonate of lead which is not very soluble. If the H₂O which has stood over night is drawn off there is little danger of Pb poisoning.

Acids or preserved fruits attack the lead in the tin or solder and you get poisoning in this way. The so called tin foil usually contains a large amt of lead. Some specimens have as much as 90%.

Paints are very liable to lead poison. due to use of white lead (carbonate of lead). Plumber-type letters are prone to this trouble. Lead is a cumulative poison and is absorbed as such. and it may not appear in the urine until after treatment. Inhalation of lead dust in the mines causes many cases of poisoning. Grinders & colars are very liable to this trouble.

Red and yellow lead are used often as a pigment and thus cause poisoning. Use of lead in enamel of cooking utensils causes many cases of poisoning.

Ordinary porcelain lined dishes
may hold Lead: in composition
Cleaning bottles with shot in
apt to cause poison. due to the
formation of a slight amt of
 $PbCO_3$ on the walls of the bottle
Cleaning champagne bottles with
shot a few shot remaining in the
bottom of the bottle and then
green due to poisoning
Ordinary red wine sometimes
contains Lead.

Coloring of food and confectionery
with chemical yellow often causes acute
poisoning due to much to chemical
acid & to Lead.

Many hair dyes have a large
amt of Pb.

Acute poison is caused usually
by ingestion of sugar of Lead.

Use of matches not lighting a
pipe with the ordinary matches
which are colored with red
Lead. In acute poison.

We get first symp of an ordin-
ary - mutant person. and a
sweet tooth. colic and consti-
pation. These last u-characteristic
of food as is tooth

Cilicg. to sever and in expectation
and is relieved by deep pressure
a old women in any other mutant
pressure increases - it

Lead is a stupefacient and
the stupor comes on either
late and lasts of 2-3 day
before death. Recovery from
Pb. is very rare - and is preceded
by an acute febrile attack
and recovery is slow.

Chronic poisoning

Mol. nutrition. Pale anemic
More or less yellowish hue to
the skin. Blue line at junction
of teeth and gum.

Emutation takes place. and
after a time according to strength
of patient.

Lead cc, paralysis articularia

Pulse has relaxed by pulse and
adon attracted so you can see
shape of spinal column, and in
thought to be new religion of milder
tongue learn and am aware with com-
plicity and pulse 30 or 40. Le Cernid
Arthralgia always ^{find} ^{movements} ^{of} ^{upper} ^{limbs}
joint and resembles rheumatism
and is distinguished by the
fact there is no inflammation. And
in fact mostly in lower extremities
Next comes the paralytic which
affects preferably the upper extremities
and affects first the ^{extremities}
and the wrist drops. And is
most common in the hand and
neck. This may come on very
rapidly though at a rule a
few days before.

Cerebral there is a most
Epileptic convulsions common.
And part to have dry and
free powder. Tolerant head
ache and. No danger comes

from sleeping in a fresh painted
room or for a lead or emar-
ned but the colics are due to
turpentine and the volatile oils.
In cases of chronic poison of
long standing we get interictal
Bright.

Testament is a acute the best anti-
dote is epsom salt. The former the
mild salt and the latter and
acts as an emetic cathartic

If any other sulphate is used
it should be followed by a
cathartic because the sulphate
after a time is absorbed & the
intestinally passes. To add in
case of chronic poison is in the
form of an abhorrence of Ph.
That is done by K.G. to a slight
extent hence the value of the
thing for it does not it once
we should get acute poisoning
of the K.G. produces colic and
never give it very slowly.

not able to stand it still
are not sufficient. In the end
after a time the patient will
be able to stand by itself.

Electrolysis - good for the paralytic
Peculiar form of lead colic is
distinctive. It is located in epigastric
region and due to pressure.

We do not find many PM
symp. - sometimes a whole lot
of flesh. Most of death has
been taken place some time before
of lead has much active action
a case you may get symp.
of urticaria

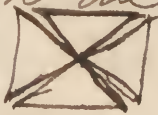
Tox. vegetation May 25. 1882

What are the symp of acute copper
poison? What is the color of the stools
How distin. between chronic copper
and lead poisoning. Pb is constipated
Cu loose. Pain in epigastrium -
relieved by Pb by pressure & Cu is
not. Pb is circumscissile, Cu diff
ure. Pb. abd is contracted, Cu
swollen and tense.

~~What~~ What is the action of H₂S on
Cu

Sulphurated hydrogen produces the
black sulphide of Cu and is de
in dilute. nitric acid H₂A

Cu is not a constituent of the animal
^{only to a very large extent} economy. It is an accumulative
poison as much as lead.



What kind of food are
most apt to remove lead? Ans.
any kind fr. acid fruit. and
many of the vegetables partic
Embarrassed. many of the alk. veg.

also remove the lead: acid
drinks. Cider. beer. Sherry wine

Mix of acetate of lead and syl
phur. make the mix of the hair
dye and oil. and give use to
most of the cerebral. symp.



Materia

-Ph. Toxicology. Rec. June 1st 82.

-HgCl₂

-HgCl₂ } distinguish between the two

-Cu²⁺

-As

-Sb

Pars green. contain both As. and Cu.
Pars green must be. forced to

Strychnine

Morphine

Either Strychnine or morphine will
be mixed with one of the metals. There
will be only one metal - the mix are
less you have pars green. Less you will
have. two metals

Oxalic acid.

*Ch.

Assessment of Cu

Zinc also an strong irritant.

ZnCl₂ acts also as an irritant and has also a corrosive action

Cases of Bi poisoning are on record they resemble an. quite strongly and they were due to impurity in the Bi and in fact all these Bi could have an. due to the fact that the Bi ore, is also an arsenical one. Prof Wood says that he never examined one a specimen of Bi without detecting the presence of as. Tellurium is often in comp with the Bi and in fact a strong odor of garlic is to be noticed which is very dist.

Lib nitrate and sub carb. combined to an acid may produce symp if irritant poison.

Chromium, this gives symp is an irritant.

When small amount is taken the HCl of the gastric juice forms an a sol. comp. and you get symp of poison.

Candy. colored. with ~~chamois~~ yellow
coloring of gums; yellow layer sur-
rounding ham - these given cases
and more probab of European countries
Coloring clothing with this has given
rise to cases of poisoning

Digitalis - and Heliotrop are both cumulative
poisons.

Poisoning by Carbolic acid. this is a
decidedly a corrosive poison.

The urine becomes dark and smoky.
and a diminution of alk sulphuric

then we have gastric symp. dyspep-
sia. in severe cases we find gut
cerebral symp and convulsions

People infected with any septic
disease are more apt to be affected
than young persons. When introduced
subcut. in the spray produce it quite
rapidly.

Test for Carbolic.

- 1st purple color - due to $\text{FeCl}_3 + \text{Fe}_2\text{Cl}_6$
- 2^d $\text{FeCl}_3 + \text{NH}_4\text{Cl}$ - color hyper chloric
or dunnish. color chloride

Dark color of urine is not in proportion to amt taken but the amt may be approxmately estimated by & amount of Sulfur. The amt carbon observed

Morphine	$C_{17}H_{19}NO_3$ (6-15%) Pure 6 me th
Codine	$C_{18}H_{21}NO_3$ (1-15%) hint impure
Herbaine	$C_{19}H_{21}NO_3$
Peperverine	$C_{20}H_{21}NO_4$
Narcotine.	$C_{22}H_{23}NO_7$ (6-8%)
Narceine	$C_{23}H_{29}NO_9$
Meconine	$C_{10}H_{10}O_8$
Opiamine	$C_{21}H_{21}NO_7$
Meconic acid	$H_2C_7HO_7$ (6-8%)

Sporifer	Capsule	Toxicology
Narceine	Herbaine	Herbaine 1
Morphine	Peperverine	Codine 2
Codine	Narcotine	Peperverine 3
	Codine	Narceine 4
	Morphine	Narcotine 6
	Narceine	Morphine 5

There are given in the order of their power - the diff cases.

From June 2. 1852.

Blue color - with Fe_2Cl_6

Froden - reagent

orange blue with HNO_3

In a legal case - do the test with Fe_2Cl_6 . and in the same sol you can do the HNO_3 test and get the deep orange.

Froden - reagent gives the most delicate test for morphine

Morphine tested with H_2SO_4 then added crystals of KNO_3 and get a flavo of color.

Principal test for meconic acid in the ferrous test. = Blood red. color which is not bleached by concave sub in HCl .

Acetic of Lead. gives the meconic acid ferrous - mag

Sublimation test for morphine - a very delicate. take a piece of Cu. put in the same. morphine. then put on a Cu. washer - and over the top part a glass. slide and you get the

crystals deposited on glass slide. This
test is good for strychnine. This test
is only applicable to pure alk. not to
the sulphates

H_2S and $K_2Cr_2O_7$ test is a
powerful test for strychnine

A very large amt of Morphia will
decolor this test, as will a large
amt of KIO_4

Strychnine sol in 2000 parts cold 2500
of hot. This is the most bitter sol known.
1 gr in 2000 parts of H_2O gives a very
bitter taste

Comp sol of I is a very good ppting
reagent for the barium series of alk.

Just try you do in a case of sus-
pected alk poisoning with conium.
test and see if you get the bitter
taste morphia does not give conium
does not adults but does in children

Strychnine should give even with
 H_2SO_4 if lucina is present you get
a light coloration

If you have strychnine mixed with
tartar emetic. you will have to separate
the strychnine by washing a little of the
powder with chloroform. and use the
fluid crop. and then do the tests
with the residue.

If. Morphine & pruric are in the same
way. alcohol. alcohol also takes up
oxalic acid and $HgCl_2$. but these
do not interfere with the tests for either
morphine and strychnine.

The chemistry of the alk is very diff
due to large amount of organic matter. and
the quant used is so small that you
do not find any undissolved in the stomach.
In Med. the dil H_2SO_4 removes all
the alk and all organic matter.

Opium poisoning

This is very common. due to ease with
which it is obtained. Pruric strychnine
Ac. $HgCl_2$ are the only 4 poisons.
but which are recognized a poison
by U S Laws.

Laudanum. resembles much that of khatul.
Powdered opium given for cough.
Laudanum and. all of morphine are the
principal sources of pain.

Pargyline is a frequent source in children.
Action of opium is a compound one.

Sleep is produced by only 3. (see list
given two pages back.)

Narcosis is more common. The morphine
and does not relieve pain.

Therms of opium if old has not much use.
cotton. Various prep of opium may be
absorbed from any surface.

If opium is absorbed slowly we get a
stimulant effect. if rapidly none.

74. June 8, 1882.

Opium.

In poison does the act of the and
opium in similar.

Acute. Due to rapid absorp. of large
doses, almost immediate, coma and
in course of 1 hr to 30 min get death
in the case pupil dilated. Rarely get
convulsions. The form is rare.

Common form. Signs come in 1 hr.
have 1st stimulant action. if absorp.
is delayed. increased pulse. ex. little
dry skin and thro. nausea vomiting
stomach. may be emption if given on an
empty stomach all the signs may be
wanting, 2^d stage anaesthesia. deep
sleep. relax. of muscles. pupil much
contracted. not react to light.

Comp. in sensibility. last for several
hrs. 6-8, 3^d Coma. deep. stupor
respirat. pupil dilated. take place
just before death. death in 5-15 hrs.

In children after get convulsions
rare in adults.

Lower - animal rest & more free.
the convolution.

Recovery - common as increase
and this kind of opium time existing
during the first stage. sometimes ^{of some} during
the 2nd.

Chronic (opium habit)

There is nothing marked off in
stage, only that a person get into
a habit. so that an immense
amount may be taken in some cases.
97 have been taken per day.

This usually results in suicide.
This habit is often induced by mother
giving this to children. They rarely
live to be more than two years old.

On the average death takes place in
12-15 hrs. in opium poisoning if
he has held over 12 hrs and has
been from stage. prognosis is
favorable.

There has been an adult has been 1 gr
of M. 4 of O. 1 dr of Laud
Children and old people are very
susceptible

Respc. suffering from B without
dyspnea and very easily affected
All cases when a much pain large
doses may be taken: P.M. congestion
of brain and some. of reflex
hemorrhage.

Apoplexy and Intoxication are the
appearances most often taken for
opium. You cannot catch either the odor
of them. Confused Consciousness
there. Pupils are diff & apoplexy
and only one side paralyzed.
Opium both in opium and stroke
you can usually partly remove.
The patient in apoplexy you cannot do so.

Stomach pump with the best
 results as good because the
 stomach is paralyzed. Pain and
 an electric battery, no good con-
 sider. atropine and belladonna
 are good to counteract the
 action. Action of atropine last
 for 24. Opium for 12. And if
 you seek to break.

alcohol you may get death from
this. After strictly with acid and
alcohol. stroke with benzoin and
get succinic. the acid alcohol
and the morphine may be taken for
the alcohol. by shaking with H₂O +
a few drops of H₂O and it is

~~Most~~ Morphine resist decay
for a very long time. and has
been detected for a after 10 mo.
Stychnine last for the same time
after of several years.

Stychnine can be detected for
death takes place so quick that
it has not been eliminated
Stychnine.

Stychnine or its emulsion are
the principal source of the poison.

Symptoms usually come on in 10 - 20
min due to great vol of stychnine
the gastric juice 1st feeling of anxiety
and terror and anguish and
by some little time the patient is
thrown into convulsion is severe
that the patient may not open.

Lead and lead. The face & eyes
etc. due to interference with cell.
The corner a bit up and the
face may become pale. Intermittent
Crisp 5-15 min. and then
another convulsion. which if the
case is to be fatal & longer. The 1st.
The 2nd is a more severe adult
death taken place in the 4th or 5th
convulsion. if death is a convulsion
it is due to obstruction of
if during interval due to 2nd phase
then death usual taken place --
about one hr. longest time a record
is 6 hrs. All that which produce a
slightly it can hinder the sleep
and the symp. all tannic acid
comp for tannic of strychnine
opium has. doses.

1/2 gr has produced death in 25 min
1/10 gr has killed children.

Treatment - to remove
strychnine and the way to do it
is to chloroform the patient.

Ether and chloroform are the remedies

Tox June 7 1882.

Tannic acid is good to give in cases of
strychnine. then give chloroform. chloro-
also gives relief, and should be in-
fused subcut. ~~cure~~

Divid.
Surface of body is rigid if death
taken place during convulsions

Fingers and toes are very strongly
flexed. The post mortem wooding is
very marked and last a very
long time. All organs are congested
some are full of matter in upper part
of cord and brain and may be
of apoplectic hemorrhage if the autopsy
is made ^{soon} after death. The heart is
empty and contracted, digestive
organs show nothing.

In children death may come
before the 5 convulsions and in
jammals with only one.

Epilepsy. Here previous history is
taken into account. We have only
one convulsion at a time.
Intellect is lost during the fit

In strychnine the mind is clear.

Tetanus then varies from strychnine by the duration and progress.

In tetanus first the jaw is affected in strychnine it is not first and in tet the convulsions are much longer and death is from 6-10 days.

Puerperal convulsions have the patient in obsol unconscious, and the intervals may be long or short and as a rule the patient does not talk intelligibly during the intermissions death rarely takes place before but I believe in strychnine it does - the patient talks also is noticed.

In puerperal convulsions the signs of nephritis is usually present but not the same in puerperal convulsions and the convulsions are longer. Strychnine occurs in the crystalline form. some is rapidly crystallized needs it slow

sol. in 2000 parts of cold. 2500 of
boiling H_2O taste a most bitter sub
known readily sol. in acids and a ppt
by alk. The salts are very sol. in H_2O
+ drop. of sol. is very bitter.

1 part of strychnine given a very bitter
taste with 4000 of H_2O , and can
be tasted in 420,000

Insol. in ether slightly sol. in chloroform
slightly sol. in amyl alcohol.

H_2SO_4 + K_4CrO_7 , detects 1000001 gr
Best test - the frog test

Brucine, not as bitter as strychnine and
acts similarly, but is not so strong.
The may be changed into brucine by
reducing agents - brucine to strychnine
by oxidation

Brucine + conc. H_2SO_4 + well 110 give an
orange colored zone at junction of
the water and acid (test for nitrate)
Shaken with benzene, removed shaken again
and allowed to evaporate and you get a
pale clear. comp. of strychnine

Belladonna datura stramonium. Hyoscyamus
niger. Most cases are accidental due
to taking the intended for others use
All are stramonium and Hyoscyamine. The
alk are common to all three. The all parts
of plant are. poison. Signs are rapid
pale throat & dilated pupil. If given
several days & more dose it acts as
an accumulative poison. After a
time get dizzy green and nausea.
Incontinence & not normal potent
urine. as though drunk. Have a dull
eruption similar to scarlet fever.
Bladder and rectum are paralyzed.
There is no force for the force may
be accumulative & then flood.
And be vomited. The urine
is not done but if any they a
little increased.
Sometimes get convulsion after a
time have a gay and wild delirium
just upon coma. convulsion and
death after a few days although
may come after a few hours

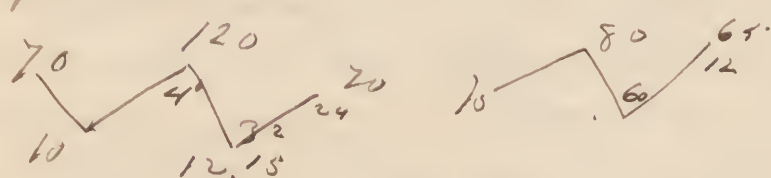
urinary, common, and after urination
 in 5. fever, after 24 hr. in con.
 give a favorable prognosis, because
 the in both limits of ur action.

60 gr of corn = 1 gr of strychnine
 Dose of bell are blue and give strong
 blue color.

360 gr of root = 1 gr -
 1/10 gr in powder form. Sub cut. .003 - .005 -
 in gr. have powder form

Strychnine pump another, purges are
 all of value, cause delay the
 absorption and in only of use to delay the
 effect. cold application to head and
 spine.

Morphine has been used to counteract
 the effects, and is quite good.



Color is increased in strychnine. In --
 and if color is livid
 both done from. No. 5 = sleep
 It = no sleep

Tox June 12 1882.

Morphine does not set well after a large dose of aporphine, and indeed - worse than nothing. Atropine after large doses of Morphia acts by contracting the blood vessels and preventing cerebral congestion. P.M. are much changed. - Aporphine or belladonnine we sometimes get congestion, and if the lungs of bell have been taken get a blue color.

Best test for atropine - The frog test - sol in alcohol and the acids

Dilation of pupil common - 10-20 mm if taken into stomach, 5-10 subcut 2-3 under eyelid. Chemical test are not of much account.

Stramonium seed are the same shape but larger and flatter than hyoscyamus - belladonna: Atropine - quite resistant

Each of these drugs have two alk. Bell. has the light and the heavy. Ardeni like - veratrum viride

This was a very common source of poison

ing has much to do with, vomiting purging
then a heart, pulse 30.

Vomiting - abt. almost immediate
stomach and enter congestion, and in
the clamping of the skin, hyperemia in
caused. And as it may be mistaken
in the urine it should be saved
except in the case of uraemia which
cannot be.

Excitement and stimulants to
bring up action of the heart.

Bayle, amygdaloidal - chloroform. all
remove the free an acid solution

A little piece of Am & H₂O₂ - first a
yellow finally a red.

piece of HCl - ^{of heart} anhydrous blue.

Alk & sugar & HCl. = 1st green then purple and
finally blue. These test = slowing of heart.

2nd - dil acids and ppt by Meyer's reagent

Legionella

Acute and chronic; today case due
to the cumulative effect of the drug.

Acute signs of ¹⁰ commencing uraemia
from 1-3 lit

It is very hot & unpleasant in the afternoon
The temperature is 70° in the shade

When the leaves are given you can see under the
microscope the kind of leaf has a long report
from one side and that is the same
H.A. is better than H.D. in staying time

Acute, both leaf from one side as far as
and there are many in each one
forming a line it is one of the most good
designs known

Sign of report of a leaf from one side
of the leaf & the leaf is always a long one
that is the part in which it has been applied

Intense headache, trembling & more of the
same nature, with vomiting, also
as a result of the same condition
happened to some of the patients
it was the 1st of the 1st of the 1st
and the 8th

1/3 of each of the 2 3 of leaf

Prof. Chamberlain says much in strength
1/10 gr. he put in 1/10 of the leaf

Enriched, Pump, then we all that
can be done

These pages are written by the
to the author. Some test in the phylogenetic
line. In the 1st of the 2nd & 3rd given by the
author. The 1st & 2nd given by the author
of the 1st & 2nd given by the author
The 1st given by the author. The 2nd given by the author
a small test of the 1st given by the author
nearly empty and the 2nd given by the author

Toxicology June 3 1882...

Prussic acid This is classed among the
narcotics.

Very active, and fatal.

Ordinary prep. is a 2% sol.

Chiefly sudorific, and a cutaneous

Death comes almost before you cover the
lips. Kely is much used for the arts
and gives many cases, raising in the
silver ships. Absorbed through the
tissues. It is made from bitter almonds
25 or 30 parts of almonds: 6 parts of anhydrous.
HCN

Bitter almond HCN = $\frac{1}{2}$ - 1 % of the acid

Cherry Laurel water has all the same.

All parts of cherry laurel are poison.

Severer of peach sweet almond.

Also. And cherry, mountain ash.

apricot, apple and pear seeds.

All alk. seeds of prussic acid are.

poison. pure form and anhydrous cyan

ide are not poison. unless com

bined with an acid.

Hg is very poisonous.

The bitter almond itself has given
no cases of poison.

Large dose. - 1-2 min 2gm com.
patients fall at once eye fixed prom-
inent staring, pupils dil and insen-
sible to light. Canthi flexed jaw re-
fractory to mouth, respiration and
limbs convulsive, odor is usually
precipitate death very quick in
a few min. usually person is
not seen before death.

If a small dose.

Symptoms in a few min. wt, pain
and dizziness in head. Nausea
at times. Loss of muscular power
inside of two minutes. Frothing
at mouth eye fixed and
prom. pupils dilated. Later at-
-onic & spasms. Case of seven
people. Time of death varied from 20
min to 45.

Small dose.

1st stage dizziness convulsive respiration
contracted and chest
2. Stage of convulsions.

convulsion not defecation.

3. Coma. relax of muscles. faintly.

The odor of prussic acid is important and should always be sought after. Tobacco may disguise the smell. The loss of this flower is very important. usual time periods of 2 min. usually one.

O from alcohol. prussic acid may all produce coma.

Prussic coma very quick. often never made of 15 min. cerebral, cord in. O may be up. P never. O pupils dil. P contracted. A = adre. P = death a few min. O - 6 - 12.

A = recovery

No chemical poisoning. Extent of surface is somewhat limited. with one blue. frozen & then sloughed. Jan ret eye from pupils dil.

O is a family the body of the intestine long and I remain bright.

Sign of blood is cherry red.

I believe benzole. the sign of blood

in brown. The odor of the two is
much alike. Smallest dose $9\frac{1}{10}$ gr.
Odor is often absent. The limit to test
the temperature dose and the fruit are
in very narrow

Fats dose 1 gr. oil of bitter almost
17 m. $2\frac{1}{2}$ gr of KEN - fruit

Death is from a few hours to 1 hr.

Cold ~~oppression~~ ^{et al} ~~emulsion~~ stomach pump
oil soc of iron ferric and ferrous
sulfate. These for. Muscular blue.

In doing the test for KEN the fruit
should always be a little old. for a
free acid under the action the acid
is a little HCl and so on.

The test gives shows $\frac{1}{2}$ 25.000 gr.

Lipsum $\frac{1}{3}$ 25.000

Ruber $\frac{1}{10}$ 25.000

Contents of stomach must be done and
fumes caught in a c of AgNO₃.

28 days in the longest time that
possible acid has been detected

5
I and Niti benzoc. diff.

Niti benzoc. Am a latent feverish which
may last for several days due to
delay in absorption. because it is
strongly absorbed. if followed up with
a dose of electricity in rapidly blistered
But still D. is much more rapid
Both produce the same group. &
character but not in blistering
with rapidity of blood. and action
of being longer last longer

Tobacco and poisoning is a mild
form this is not uncommon to nausea
vomiting.

Mertins and Corcor produce diff
group. Mertins produces immediate an-
concomitant and death and it
is so rapidly absorbed that it does
not reach the stomach it is strong
alk and corrodes the tissue and can
often be detected in them. Ther-
~~as~~ sub is very etc.

San Louis. poison cases are usually
accidental, due to rapid absorption of
the alk through the skin

Washing clothes around the skin
for contact with cases. sym.
Circum 5-6 mm. dizzy, rest pain
in stomach. vomiting. paroxysms
of spasms. slow respiration, death
in from 10-15 min. rarely jumping
same symptoms are produced by absorption
through skin, diff. part of body
has been 2-8 sq. ft.

12 gr. of Kover has proved good

1 gr. of alk. is given

Emetics, stomach pump.
Stimulents are given up. a good

Order is vomits after death
and this is the point. Myer re-
gards it

Hemlock poison is usually accidental
The first hand clasp resembles order
of poisoning

The prep of this way is strength
But is less than to be dizzy. For when
the heart and respiratory muscles are
affected, and as the rest is most
beneficial other muscles are affected
for the whole is though doing
as the heart is closed, emphysema
may occur. It may even lead to
from 1-3 hrs, artificial respiration
and stimulation, turning of the
body and change of position and
rest.

Exuding into the surrounding tissue
of purulent & ~~glue~~ an abscess
which forms at some point of
length it may become ~~aggravated~~
~~to a~~ abscess. Here also we
keep up our plan of Stensen
of drainage for which discharge
purulent material we feel here
to do with a case of cancer and
- such a case only the way
get an arrest of disease and
a new growth of time. & that
long anchylosis & we may
get a destruction of head of time
and perforation of acetabulum
which may cause dangerous pelvic
disease. In such a case the
treatment recommended by
Sagge - Excision. Which should
be done as soon as the disease
cases has been diagnosed
and if performed early I think
that it gives the best chance of
Success.

The one thing which.. does more.
than anything else with great success
- to stop joint disease - & favor recovery
to reduce suffering and remove
irritation - extension. This however
separates the inflamed surface and
also gives fixation to the joint.

Dr Bradford gives three uses.
1st to which it is put - 1st as a means
of overcoming the muscles. which
are spasmodically contracted
about the diseased joint. 2nd as a
means of fixation of the joint
3d. To separate the bones forming
the joint. Most extension cures
pain - many cases - undoubted
but that it fails - many more.
- equally true. Which fact -
Explained. it think of Dr Bradford
when he says that we get the
relief from pain - in these cases.
where - we have inflamed bones.
& deal with and we do not
get it - these cases where the pain

comes from an inflamed and
distended capsule. A very
common method of extirpation
is that known as the height and
fully method and it is not
only well known but quite
popular. Indeed in many cases
it has been given a credit
credit. Dr. Bradford says in
regard to the nature of extension
that if one visit the wards
of a hospital at midnight he
will find the patients lying
a all sorts of positions and
turning over on their faces
and their legs being flexed in
such a manner that the extension
is removed from the hip to the
knee. It is however the great
agent when applied with
care. It is when applied
in the manner one. virtue. it
confirms the patient to the bed
and in those cases where it

Learn - so severe as to prevent all
movements. not severe. its purpose
quite well. Yet taken as a whole.
whole without any regulating ap-
paratus - is as a means of ex-
tension imperfect. even using
it - incapable of preventing the
parallelism of the limbs. and
you potent may receive with
a deformity which - on load or
on disputation. But combined
with thorough means of fixation
it - a most excellent agent.

What has been said of this
- also true of the physiological
method of Hutchinson.

State in which the extension
- furnished by the act of the leg.
The patient being put in position
and a high shoe. Here in
young children we have a chance
of all sorts of accidents which
may render the old or keep
up the present inflammation.

They will flin step - the foot
parts - the inflammation of a sore
acute form and as the act does
not produce much pain -
the will flex the knee and have
the affected limb. Then lay
ing the wt - the diseased joint
they may add to respect the
affliction to patients who have arrived
at the stage of decumens - walking
but the limb. Then causing motion
of the joint - ~~that~~ that they are
liable to see signs of excruciation
and when setting the extension
to a great extent removed
and go yet - almost every
act a slight movement of
the joint

Thomas of Lymington mentions
a splint which gives the perfect
fixation and at the same
time prevents of locomotion
It consists of two L shaped
splints. The outer piece being so

lent on to Enclave L chest. and
on it - usually made there are
two cross pieces - the upper coming
just below the angle of scapulae
and the lower - just above crest of
ileum - These are covered with
a strong covering which extends from
one to the other. and either goes
around the chest or - passes
while the long arm - follows to the
curvature of the leg to which it -
firmly clings.

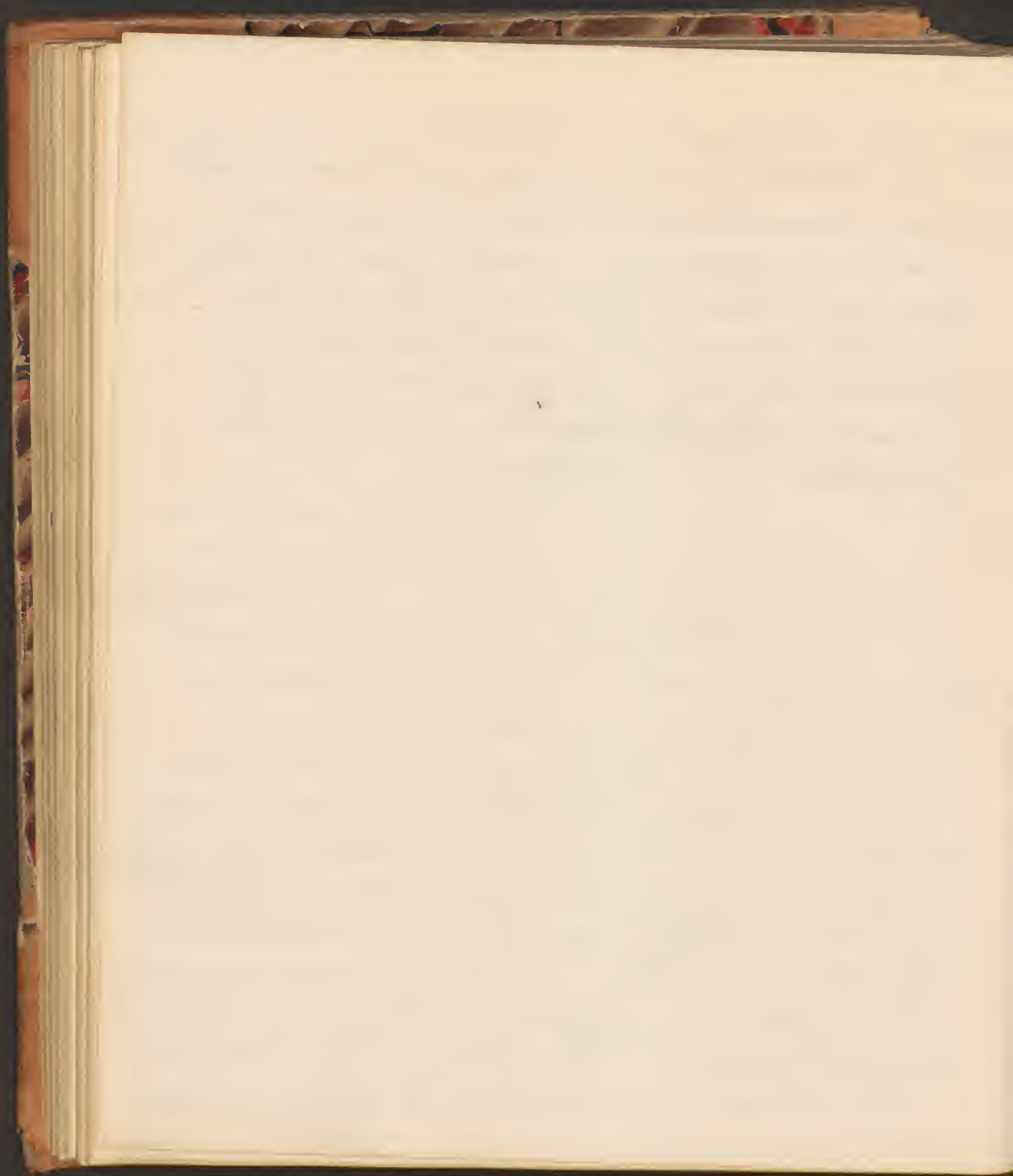
They form long & thin plates
- much used and - of great service
particulary as a means of Extension
and protecting the joint from
compression. and keeping the inflamed
surface apart: It is also to a great
extent a very good means of fixation
of the joint. but - the flap is
to hold only a 2 day position.

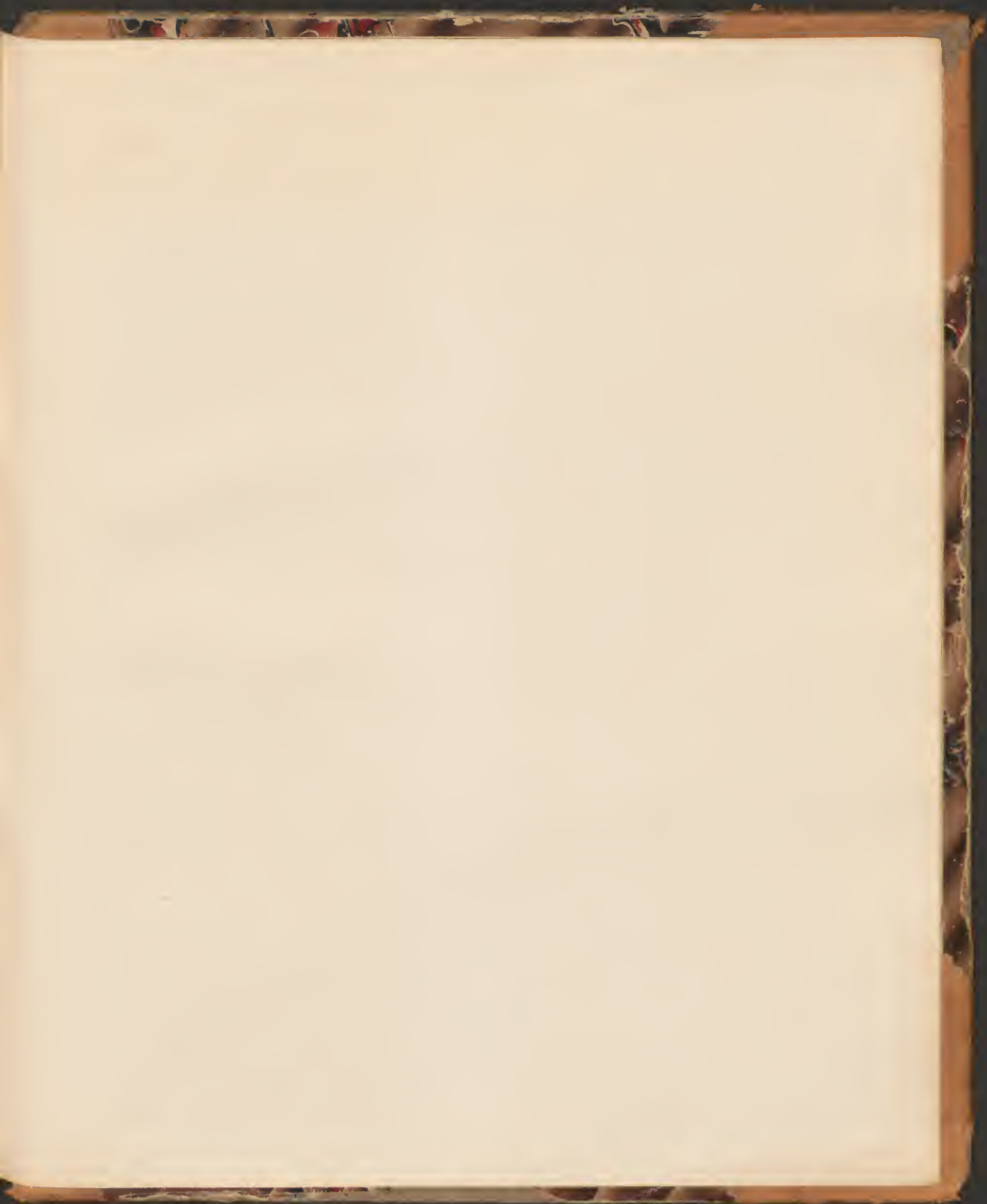
It does not prevent turning
- sleep on the station of the foot
and. it also has the ~~very~~
inconvenience of being put on with

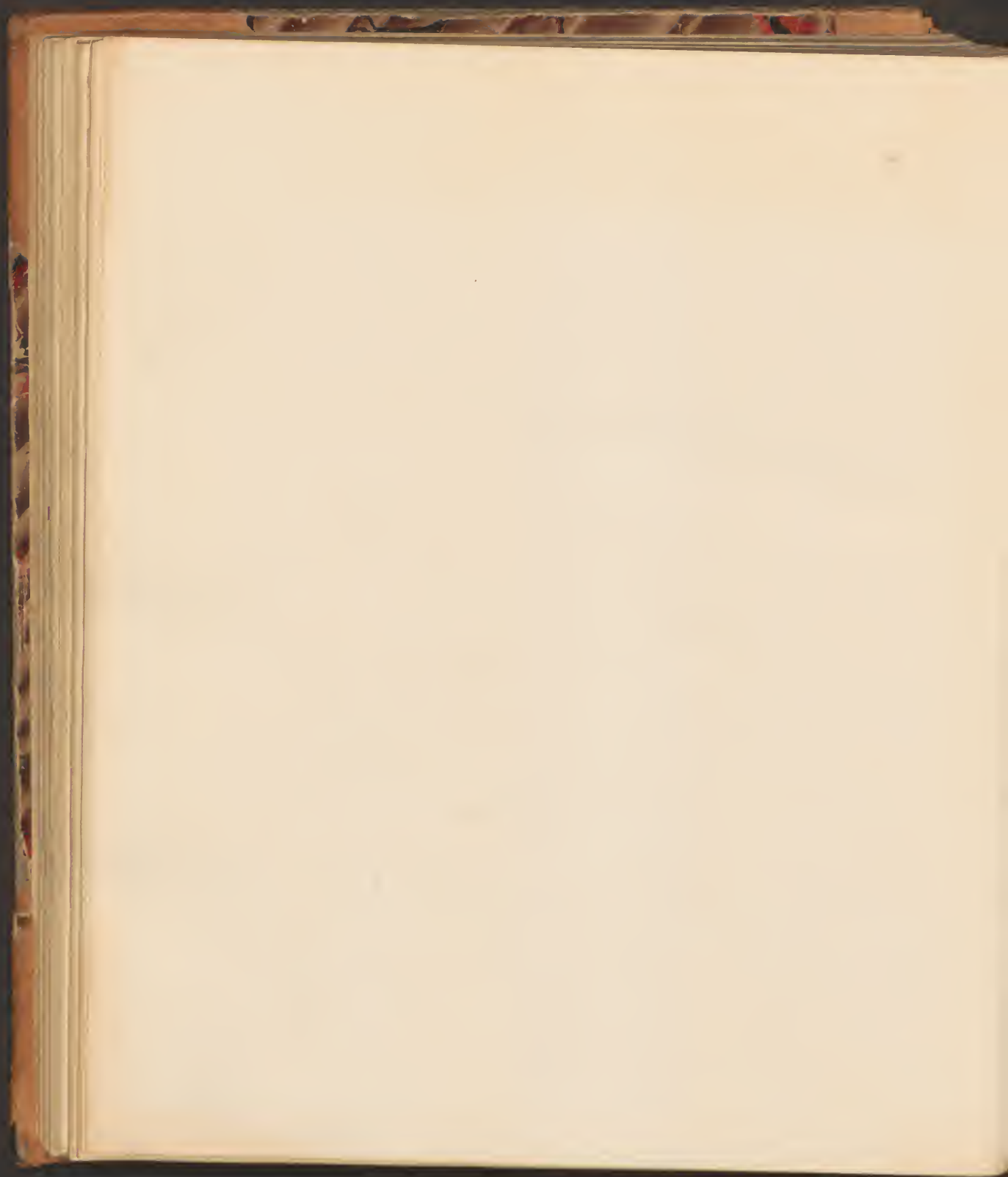
a plaster and - some about honey
But it does preserve the
hardness of the timber and
although it does permit of very
slight motion. The joint swells
up constantly - & extending to
bone - it tends to keep the
inflamed surface apart and
according to some writers
~~not~~ (not) motion under the
circumstances - not quite
impossible. After all - painful
cure - absolute fixation -
and rest either - will succeed
- - a splint similar to the
one invented by Dr. Wilson of
this city - a treatment we
are forced to adopt and one
which - sanctioned by good
authorities Under the treatment
pain will diminish - and the
child often gains in flesh
and strength. But we should
scurry & get him out with the

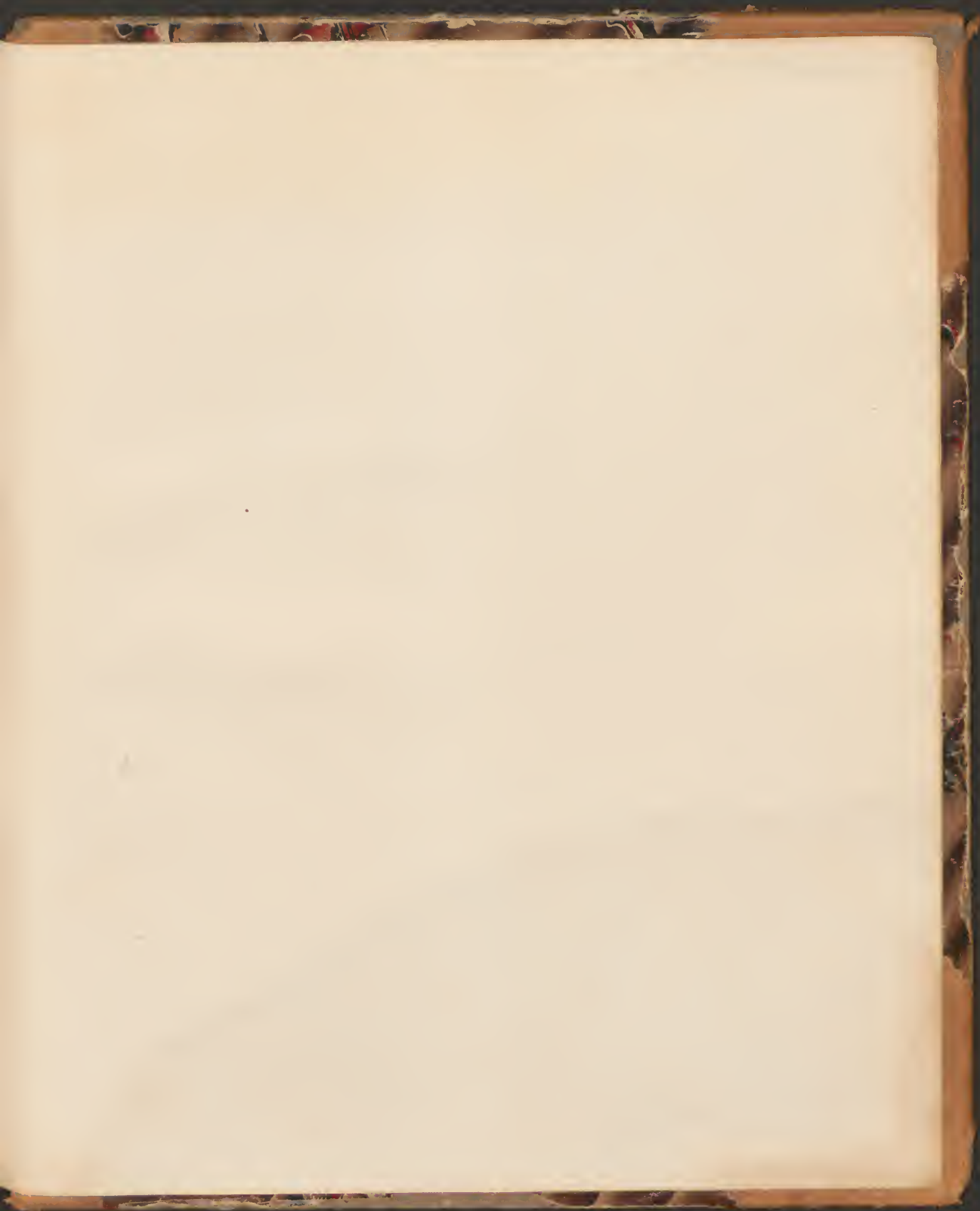
Am a son a forraller

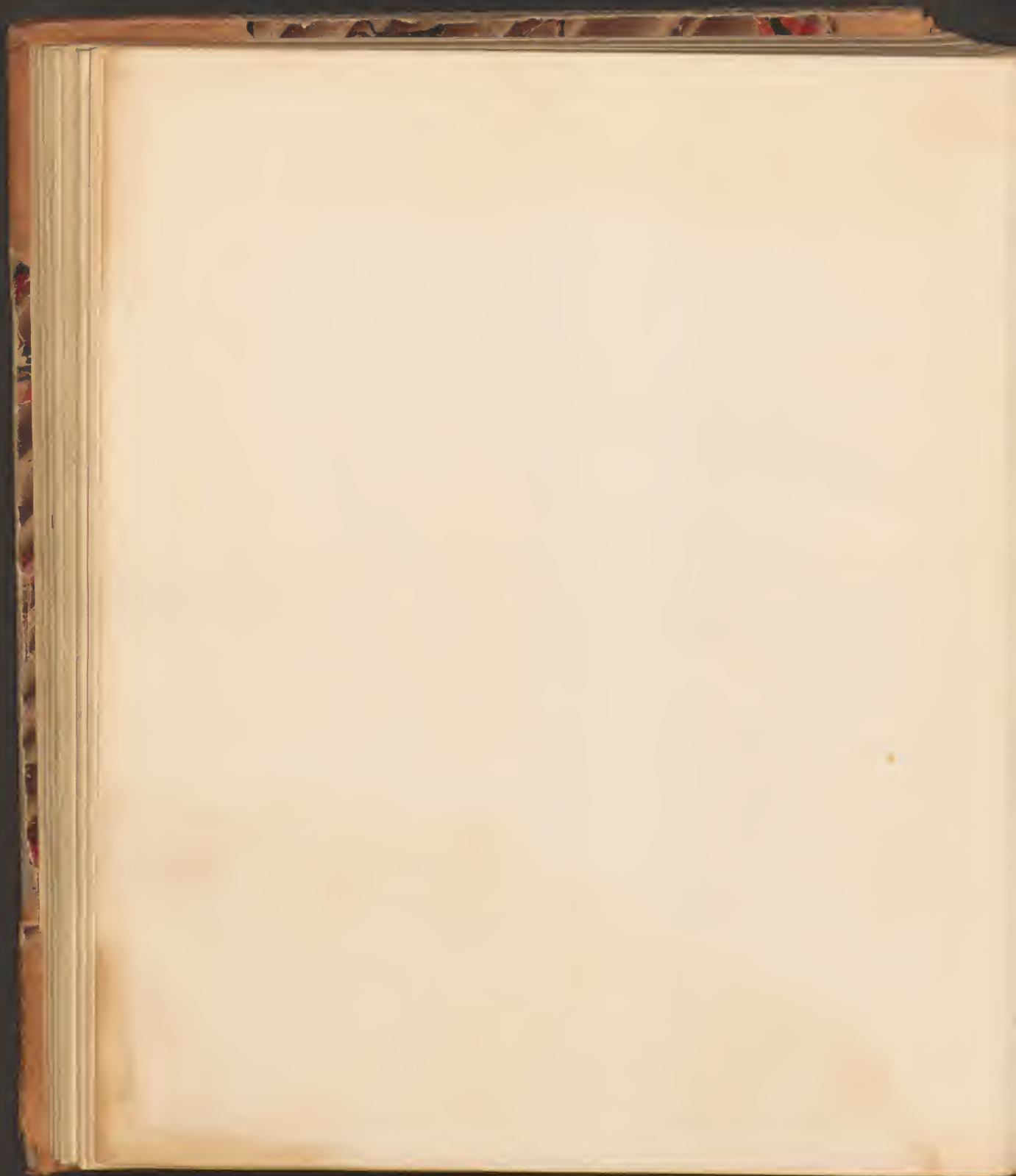
In all the treatment of all cases we
have to exercise much judgment
and no definite rule can be can-
died of the treatment of the disease.
The art must be dictated by the
experience and judgment of the
surgeon Dr Brodhead sums the
matter up as follows:-













Lotus hydangyri fava = 1 of H₂O to 300 of Ca H₂O
" " magna 1 of H₂Cl " 60 " "

LEONARD WOOD'S NOTEBOOK AT HARVARD MEDICAL SCHOOL

224. (MEDICINE) WOOD, LEONARD. Manuscript Notebook, containing Notes of Lectures at the Harvard Medical School, where he was a student, covering the period from December, 1881 to June, 1882. Sm 4to, half leather bound notebook, covers rubbed, 200 pages of Manuscript.

Leonard Wood was one of the most prominent doctors in American History: as an Army Surgeon, he won the Congressional Medal in the Apache Campaigns; a close friend of Theodore Roosevelt, he was Colonel of the Rough Riders in Cuba and later became military governor of Cuba; a Major General in the Army, he was Chief of Staff in 1910; beaten by Harding in the contest for the presidential nomination in 1920, Wood ended his career as Governor General of the Philippines.

135
133

